

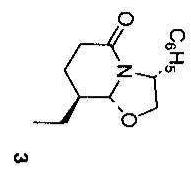
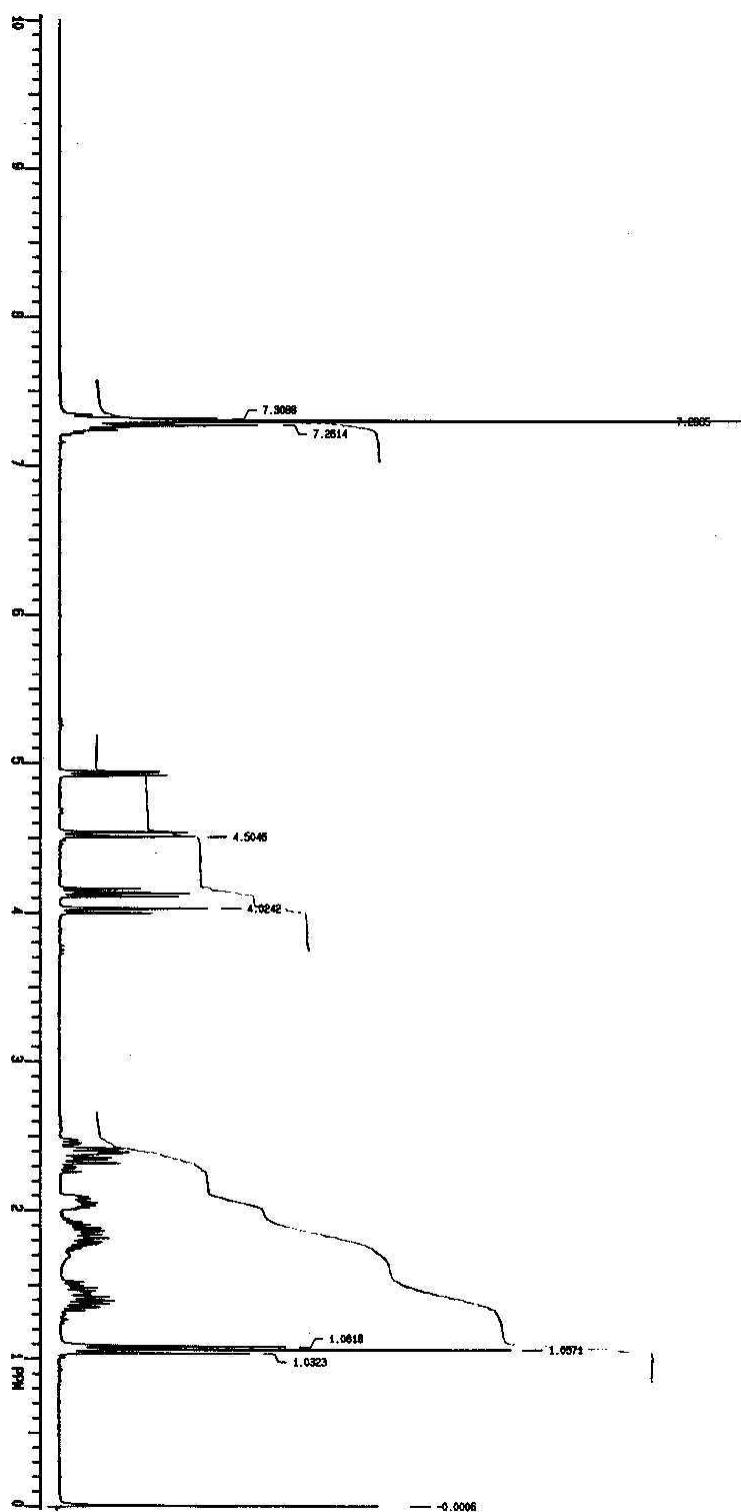
Supporting Information

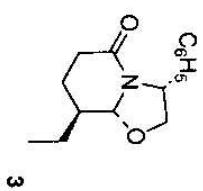
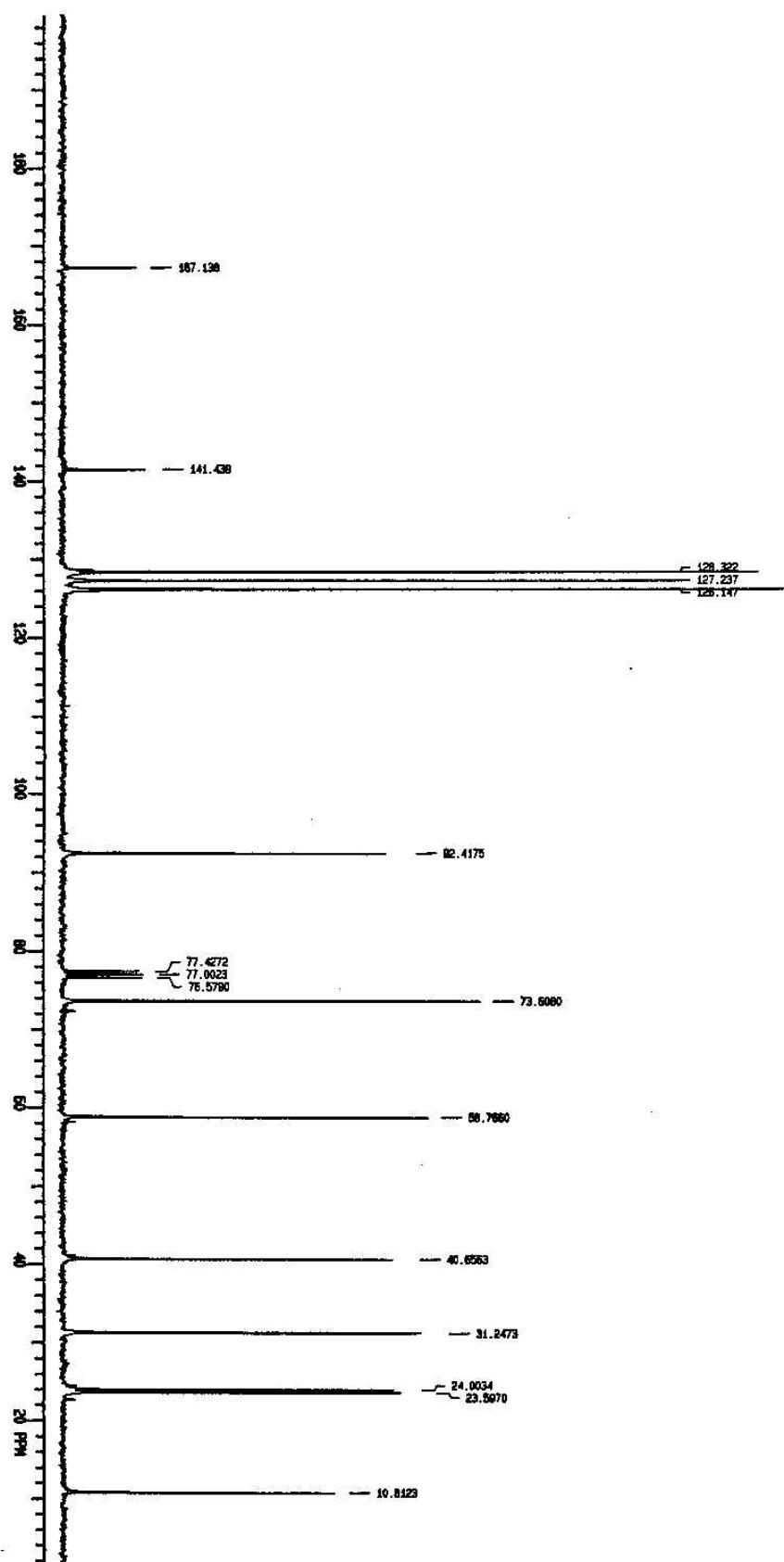
General procedure for the preparation of **6a,b** and **7a,b**:

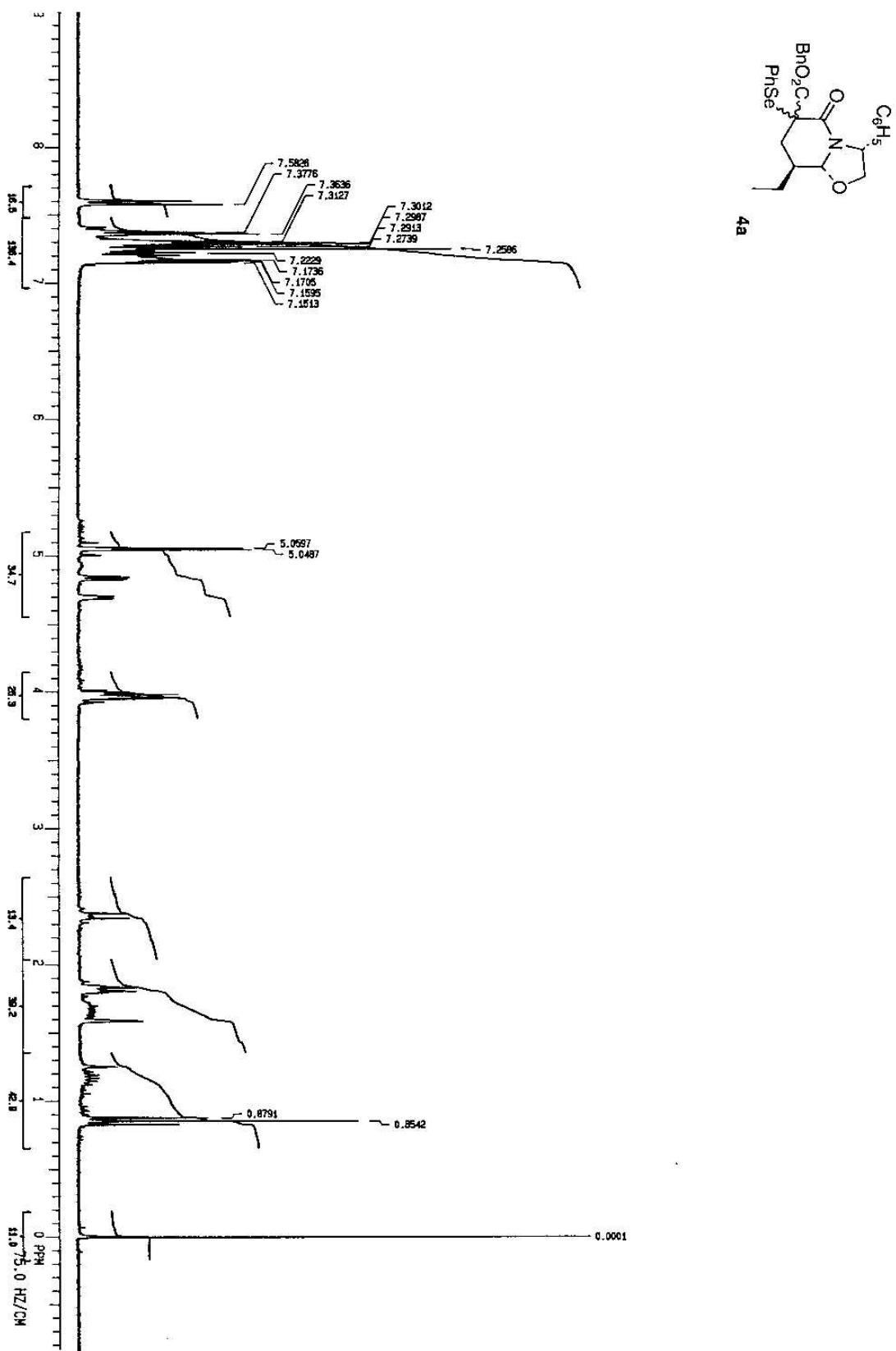
Lithium bis(trimethylsilyl)amide (2.2 eq of a 1.0 M solution in THF) was slowly added at -78°C to a 0.07 M solution of lactam **3** (4 mmol) in anhydrous THF, and the resulting mixture was stirred for 1 h. Then, methyl or benzyl chloroformate (1.0 eq) and, after 30 min of continuous stirring at -78°C , a 0.7 M solution of PhSeCl (1.4 eq) in THF, were sequentially added. The resulting mixture was stirred for 50 min and poured into 1 N HCl. The aqueous layer was extracted with AcOEt, and the combined organic extracts were dried and concentrated. Flash chromatography (AcOEt:hexane 3:7) of the crude mixture afforded compounds **4a** (86%) or **4b** (89%) as mixtures of C-6 epimers.

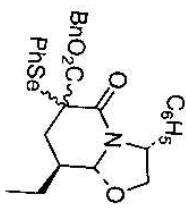
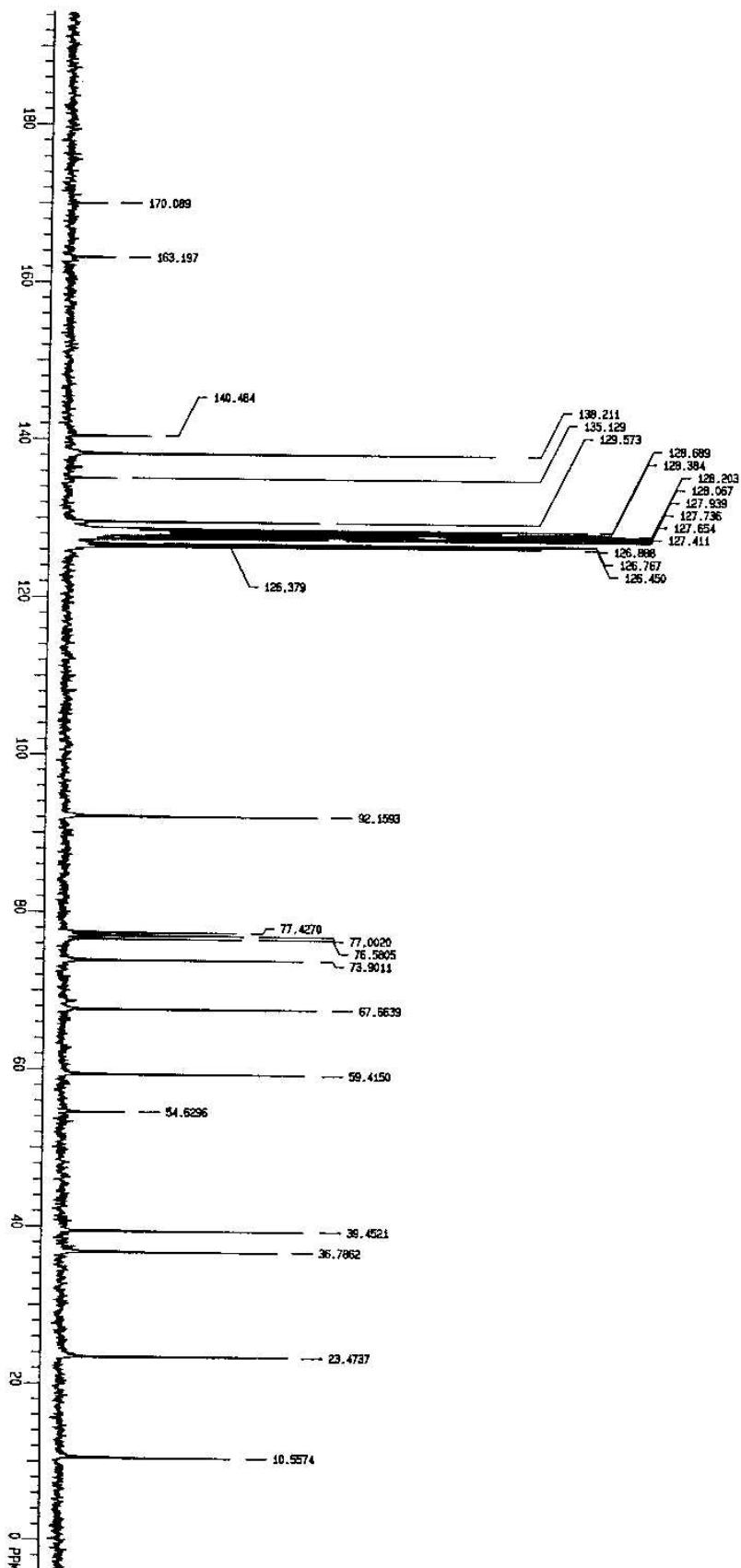
A stream of ozone gas was bubbled through a cooled (-78°C) 0.05 M solution of selenides **4a** or **4b** in anhydrous CH_2Cl_2 until it turned pale blue. The solution was purged with O_2 , and the temperature was slowly raised to 25°C . After 30 min of stirring, the mixture was poured into brine, and the aqueous layer was extracted with CH_2Cl_2 . The combined organic extracts were dried and concentrated under reduced pressure (external temperature 25°C) to give **8a** or **8b** as oils, which were used in the next reaction without further purification.

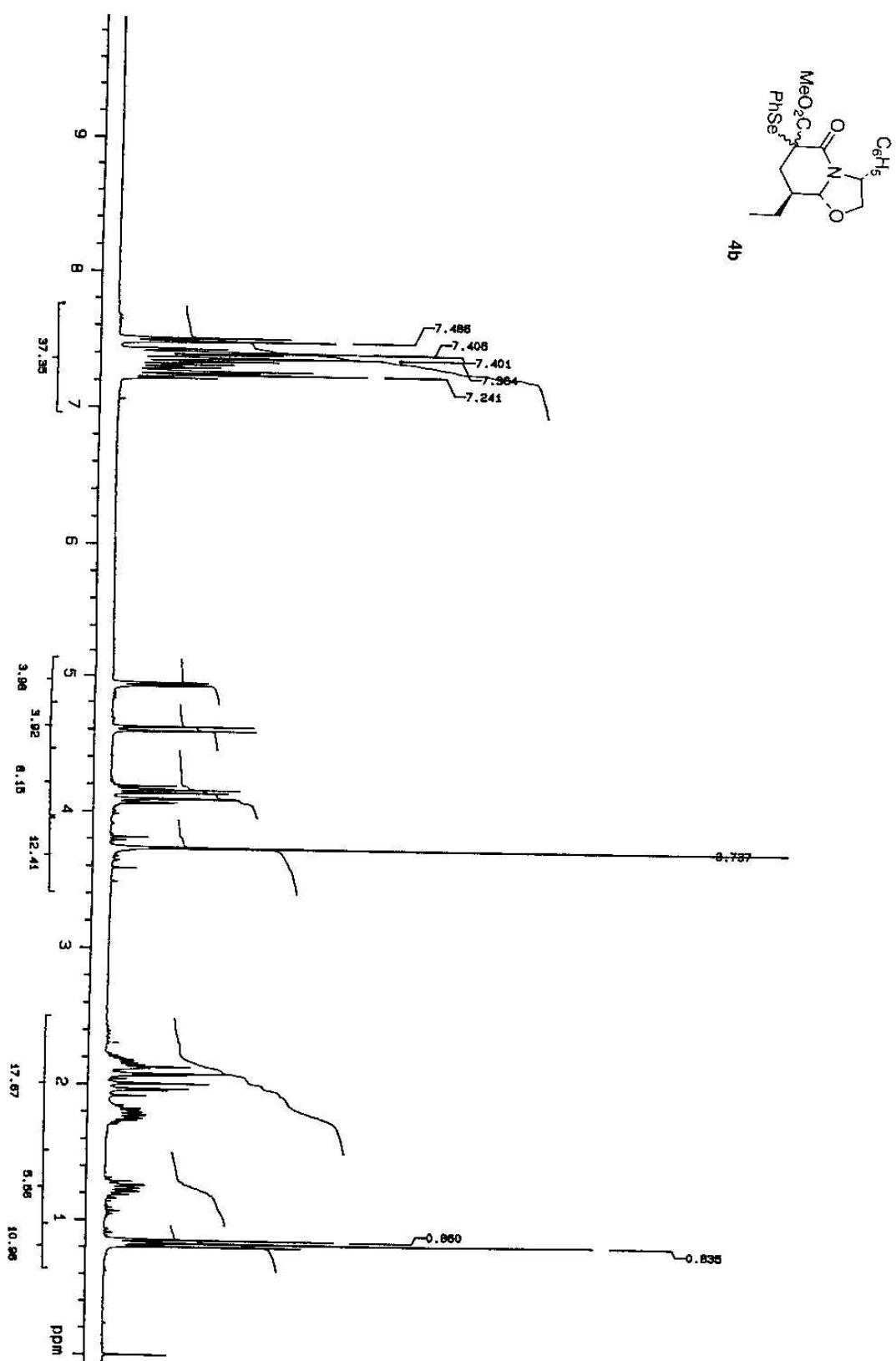
A 1.5 M solution of the above crude unsaturated lactams **8a** or **8b** (1.0 eq) in anhydrous THF was added dropwise at -78°C to a 0.08 M solution of $\text{MeCu}(\text{CN})\text{Li}$ or $(\text{C}_6\text{H}_5)\text{Cu}(\text{CN})\text{Li}$ (5 eq) in THF, and the resulting solution was stirred for 7 h (methyl series) or 18 h (phenyl series). The mixture was allowed to reach 25°C , poured into saturated aqueous NH_4Cl , and extracted with AcOEt. The organic layer was washed with saturated aqueous NaHCO_3 , and the combined organic extracts were dried and concentrated. The residue was purified by flash chromatography (eluents: AcOEt:hexane gradient from 1:4 to 3:7). **6a** (38% from **4a**); **7a** (80% from **4a**); **6b** (40% from **4b**); **7b** (75% from **4b**).

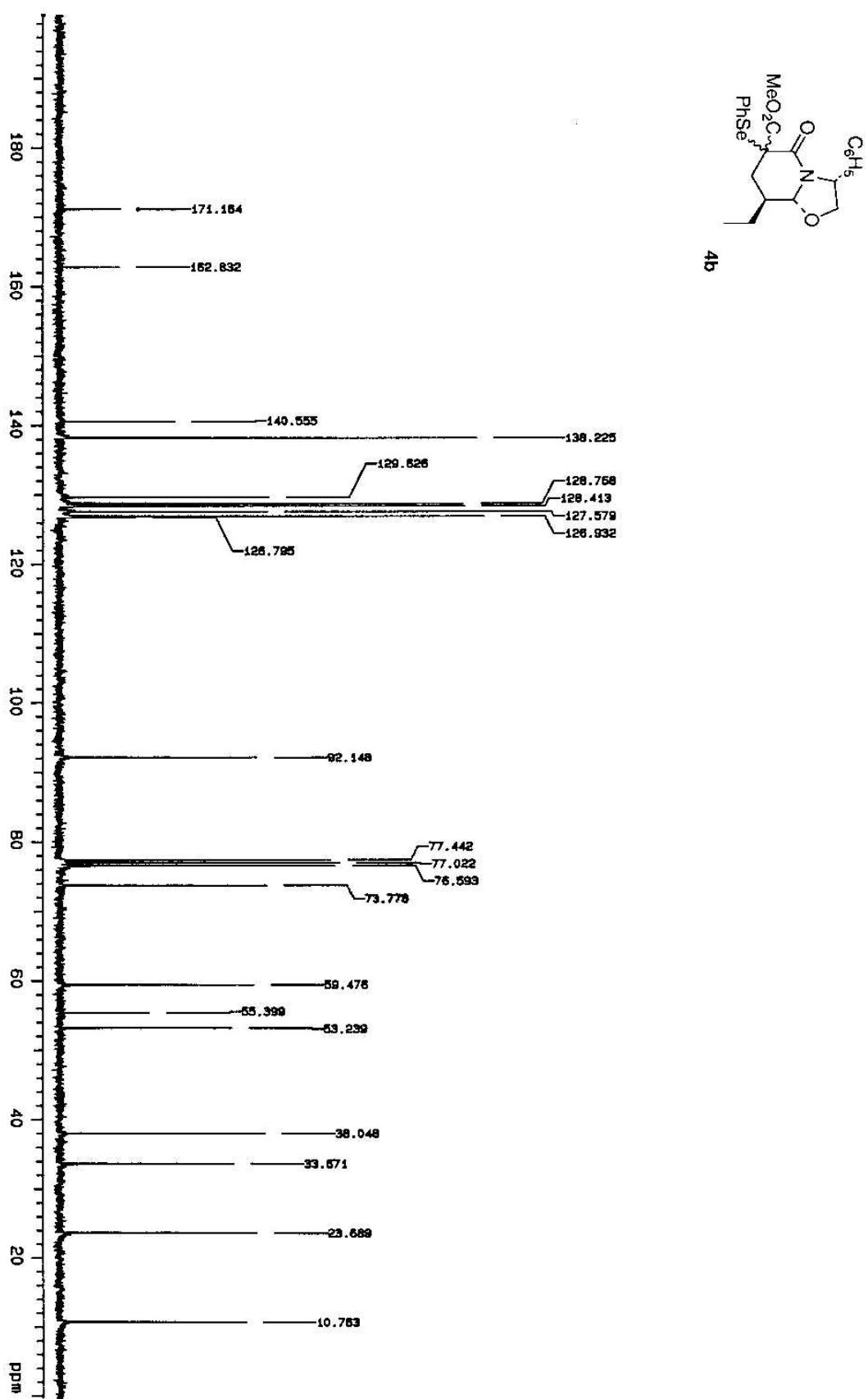


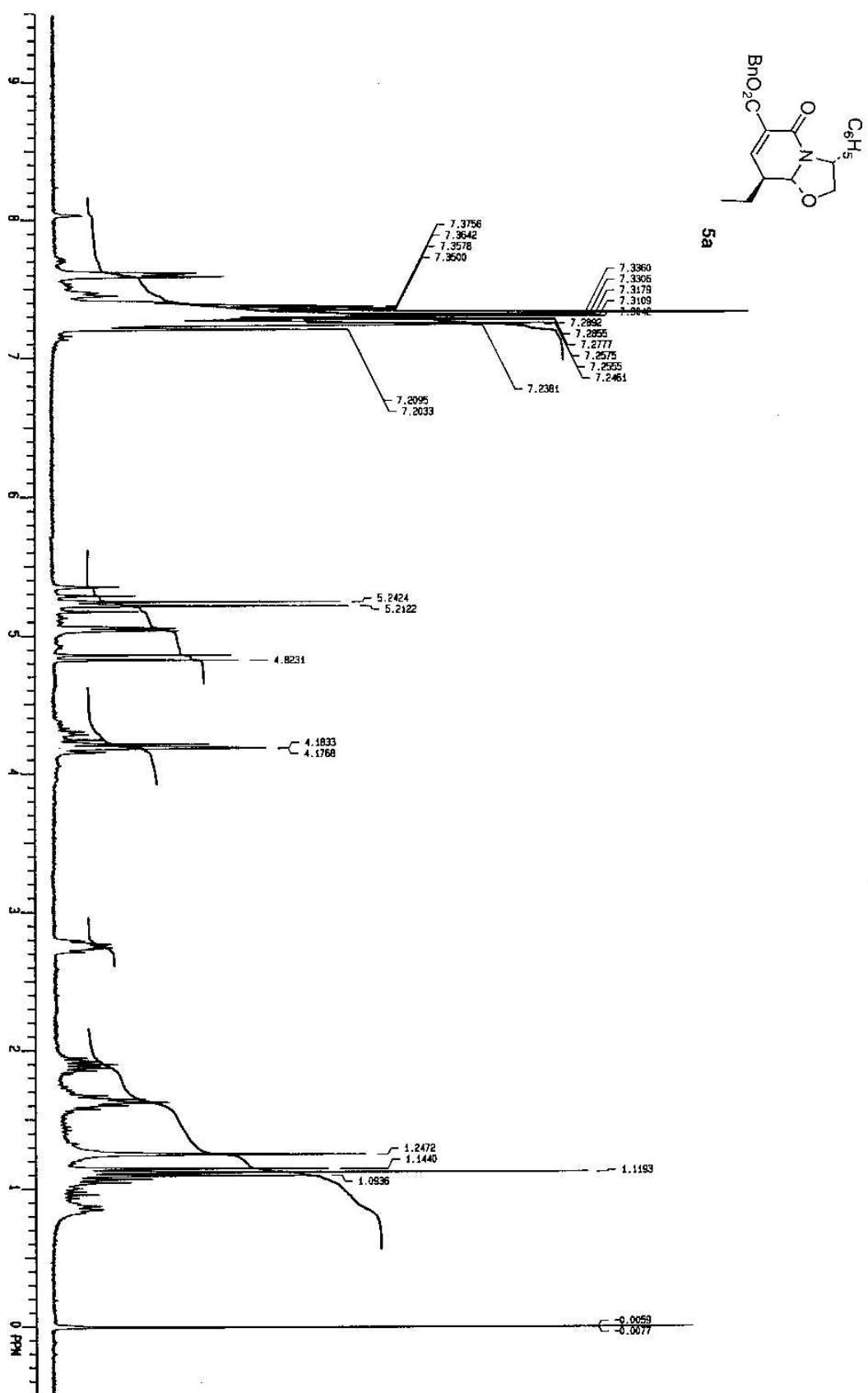


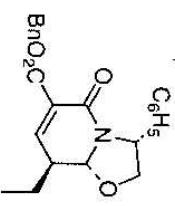
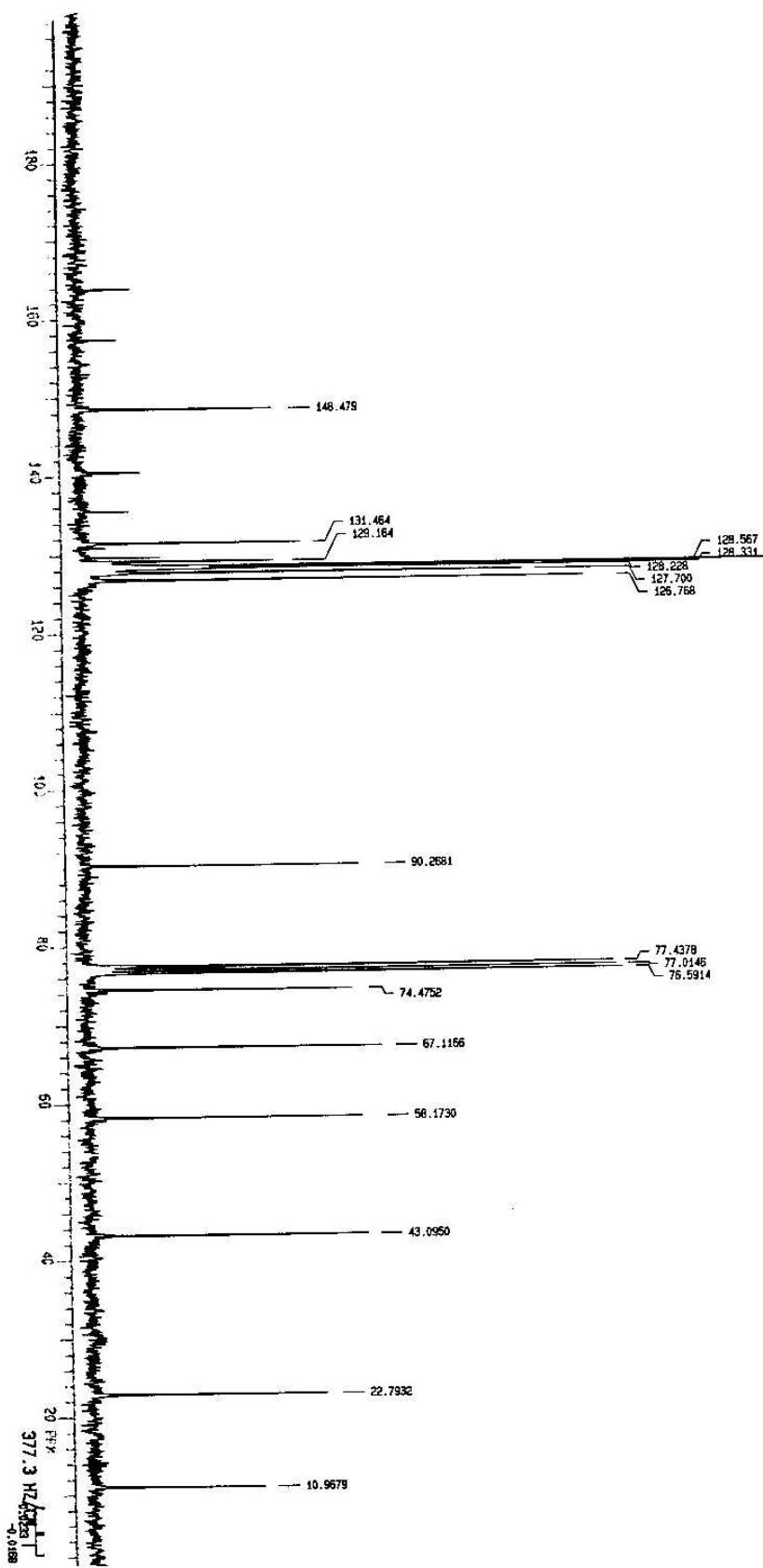




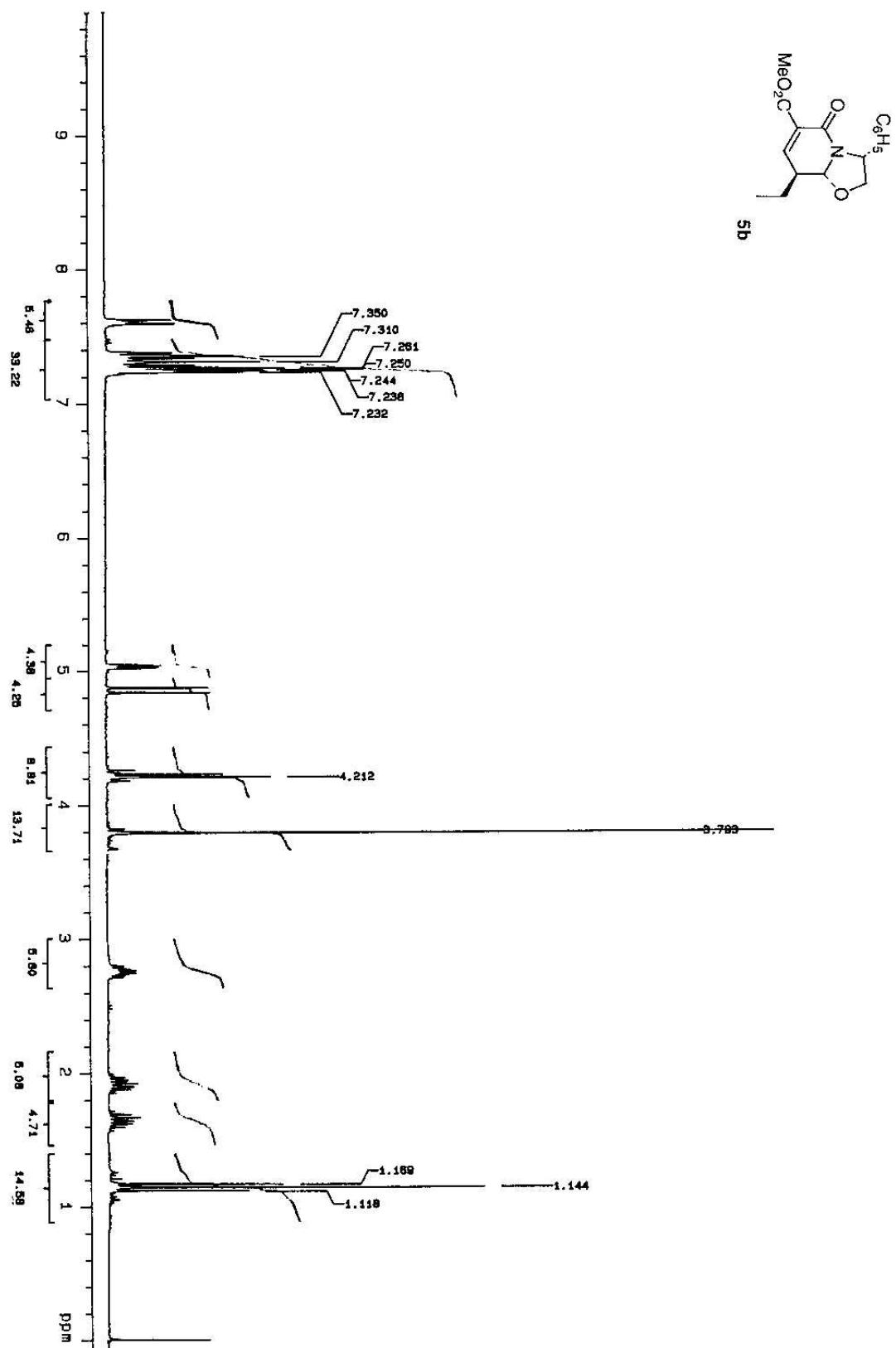


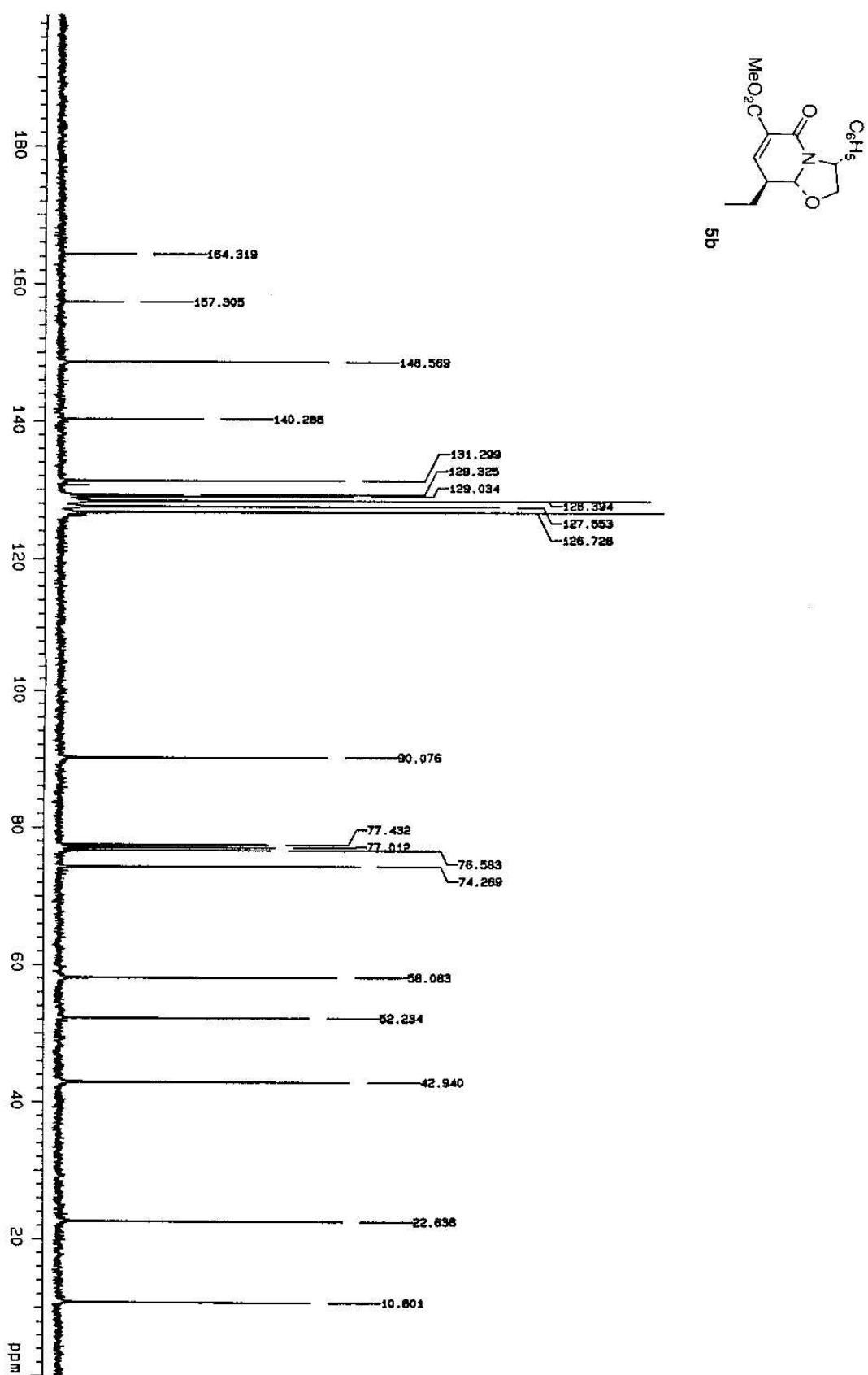


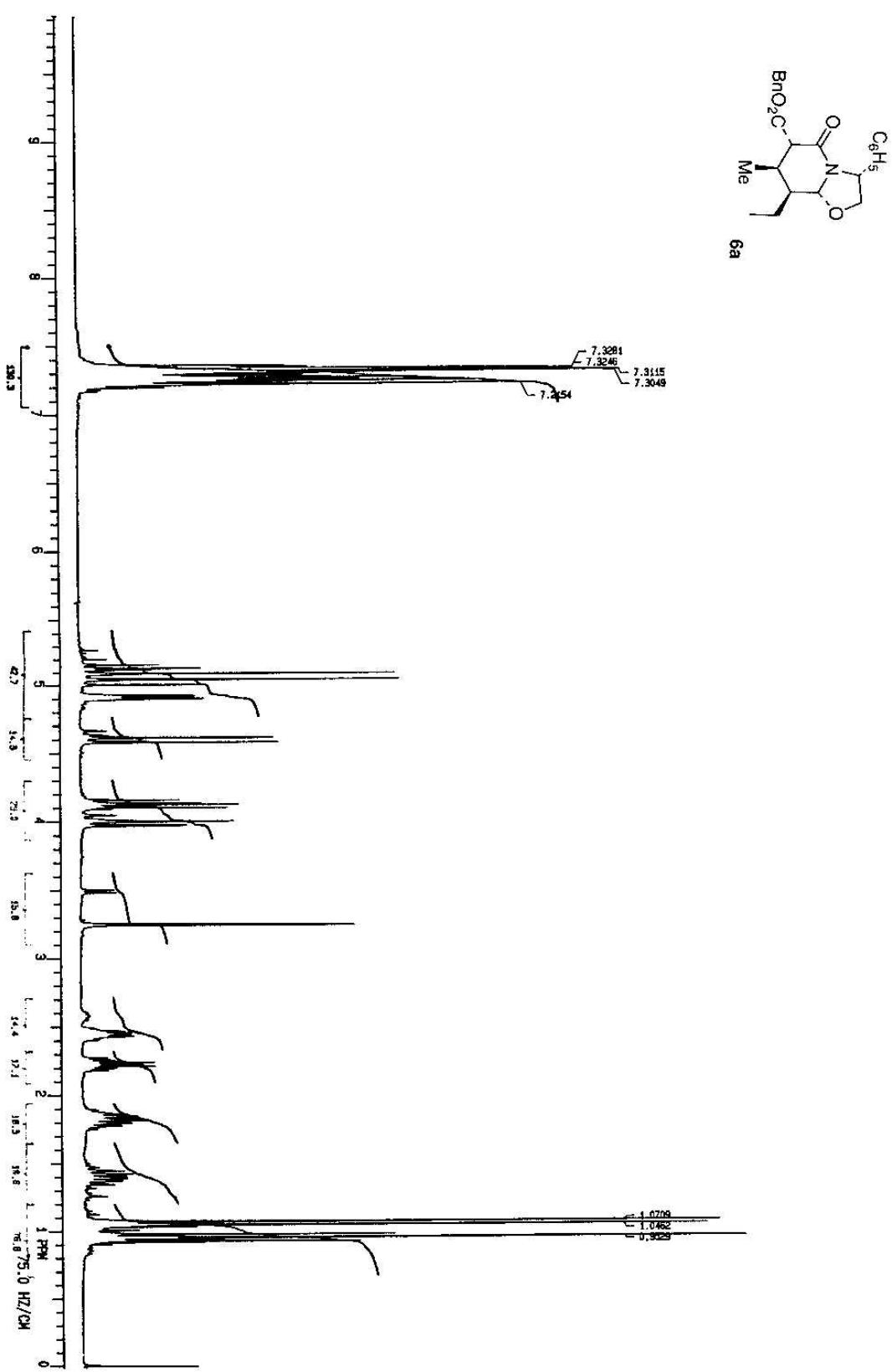


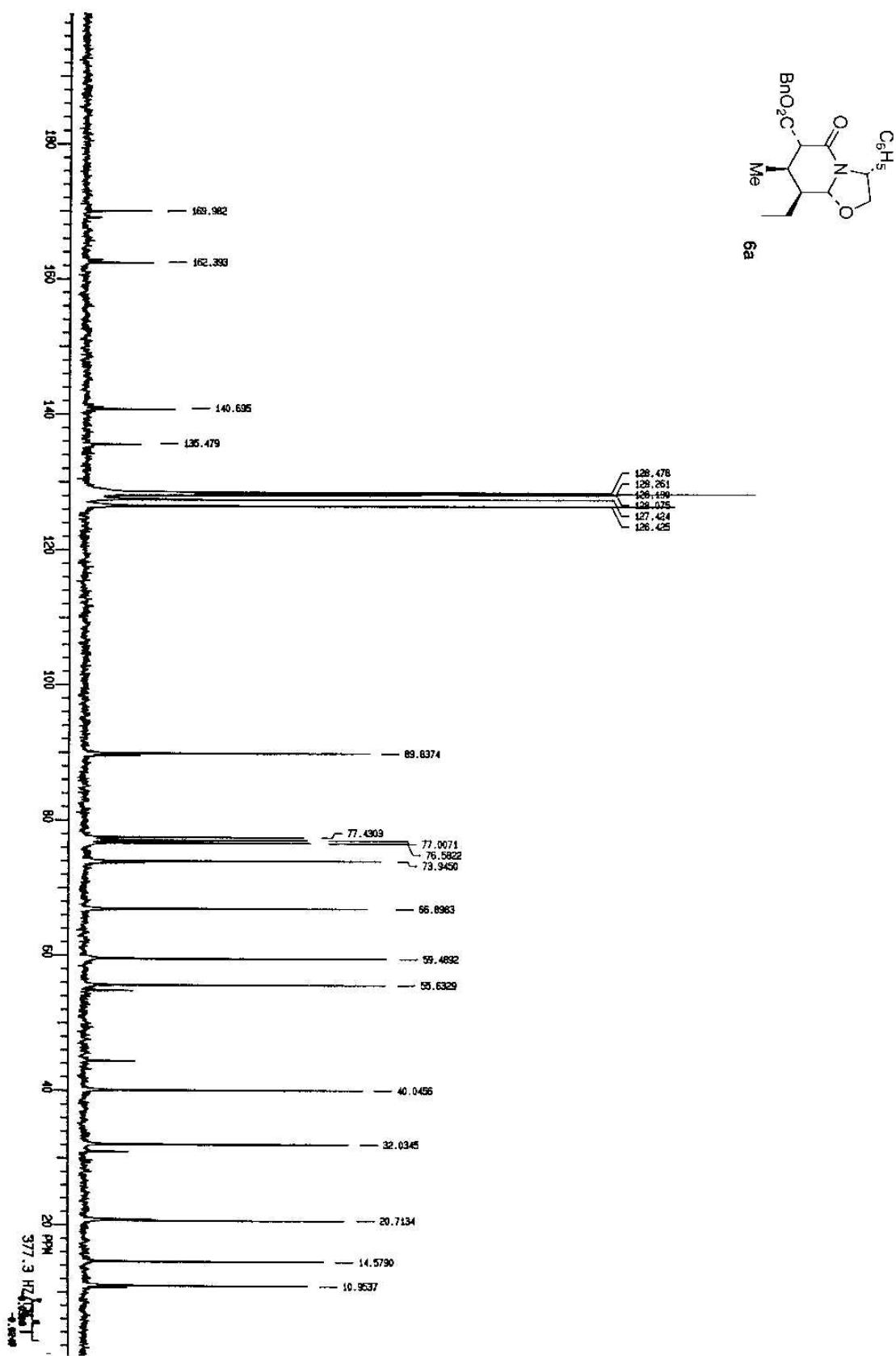


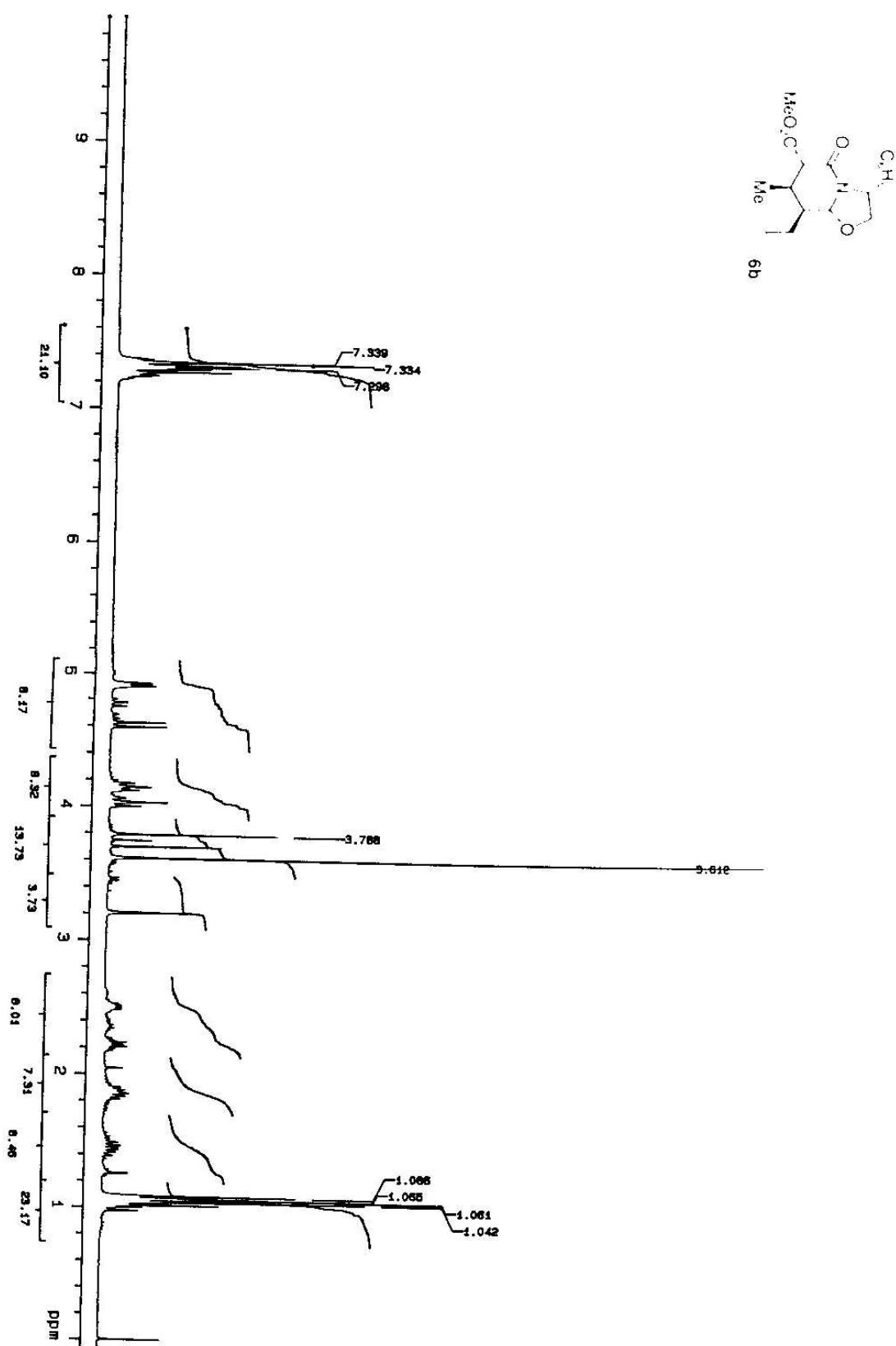
5a

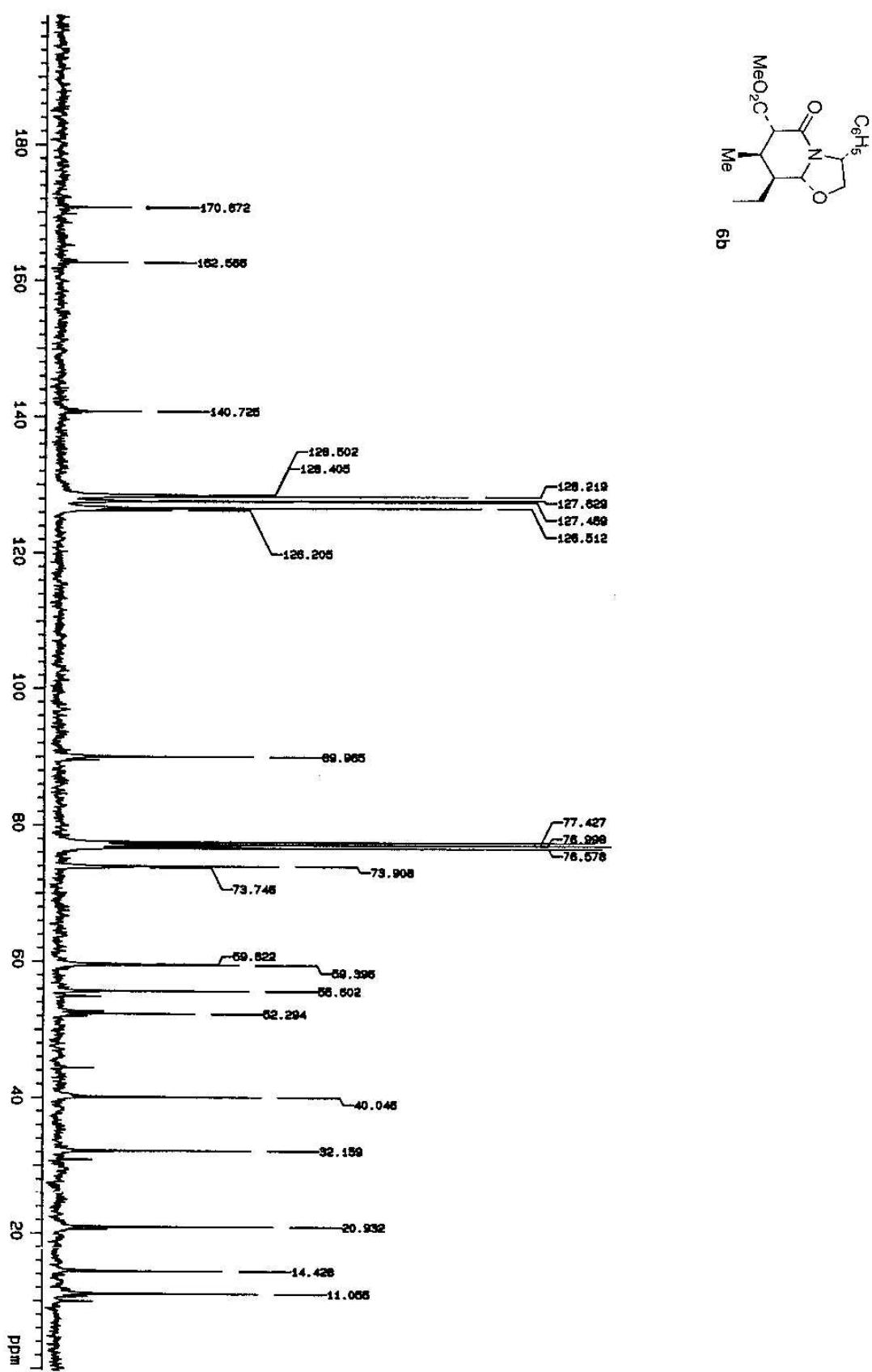


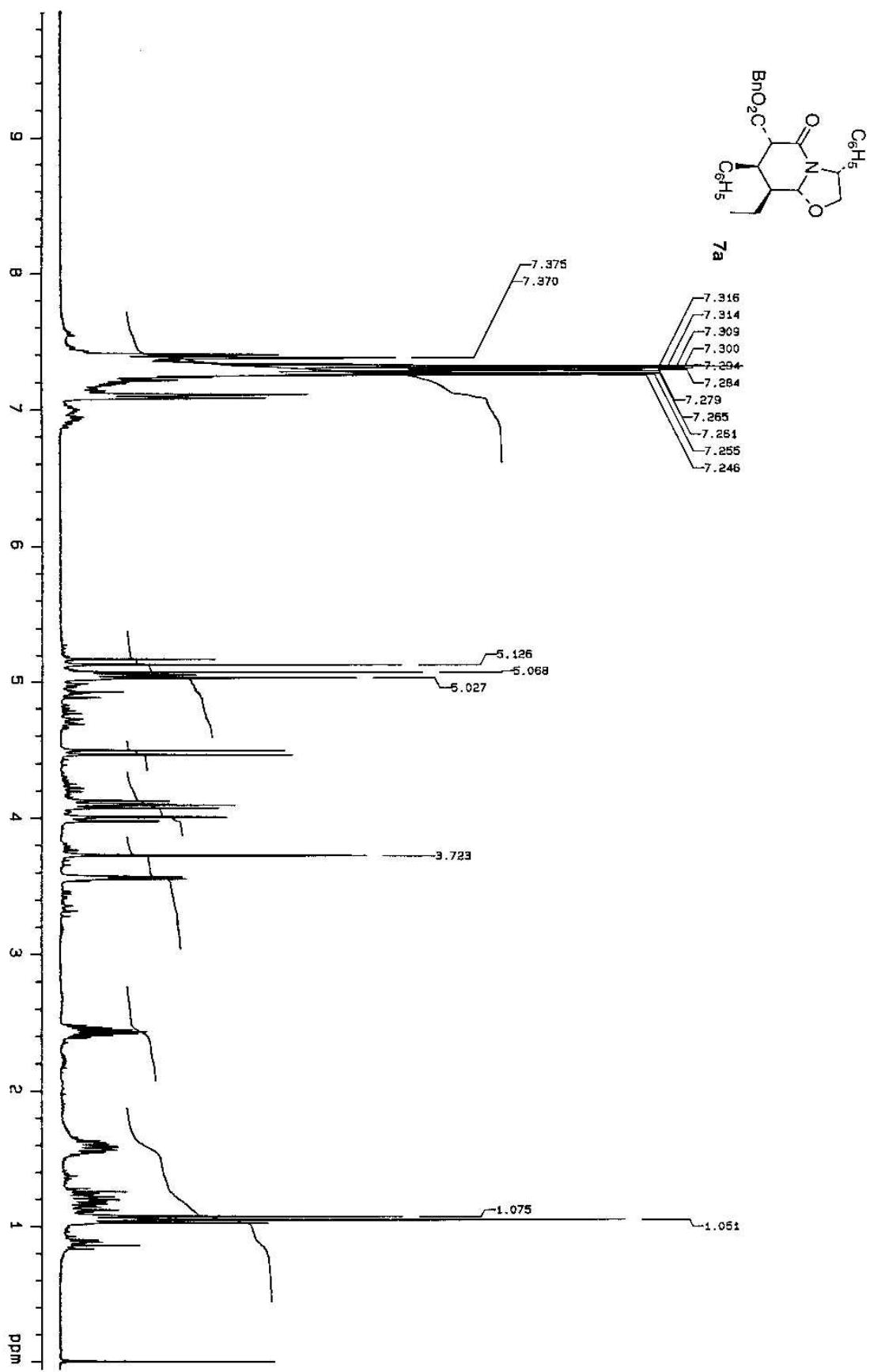


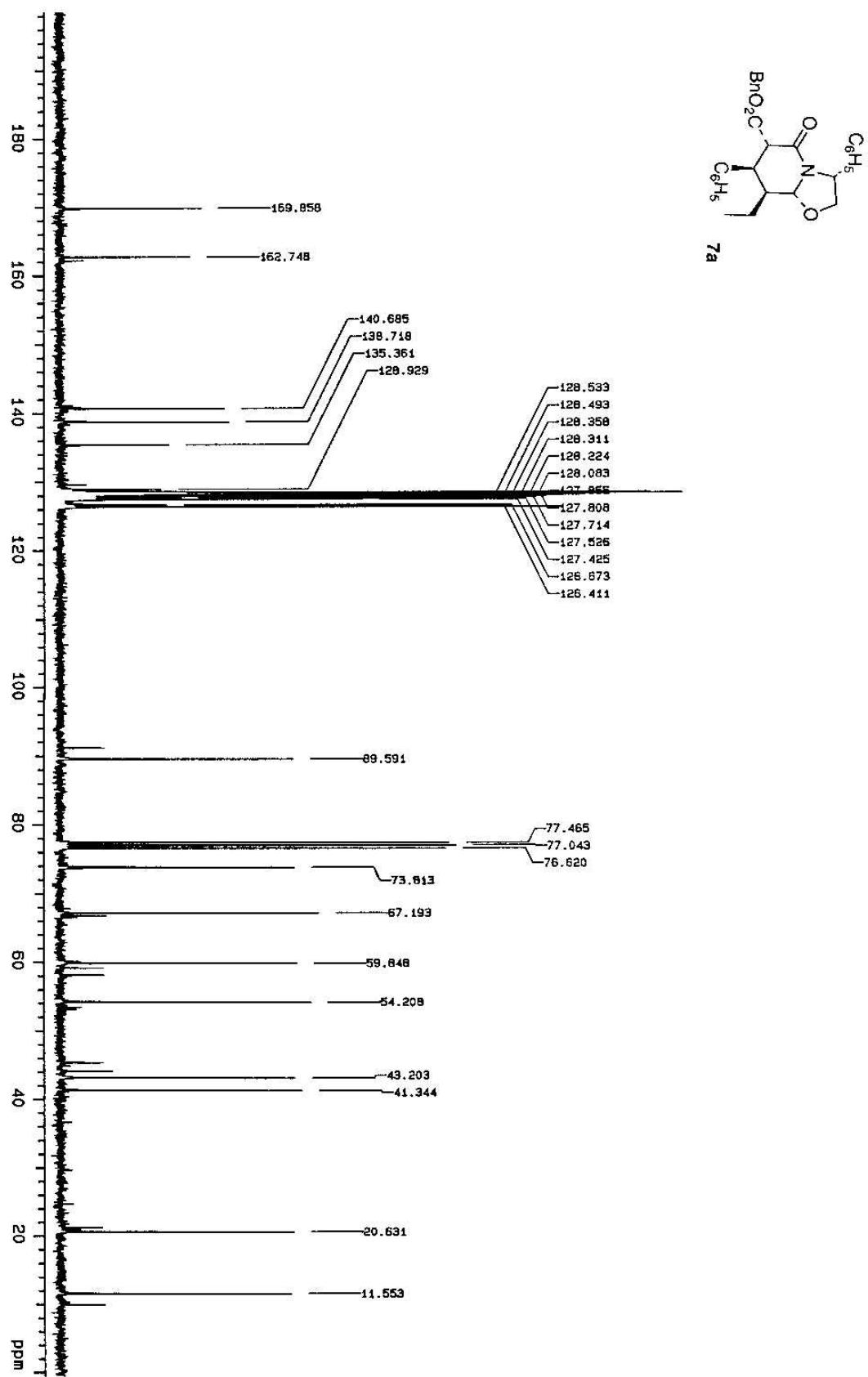


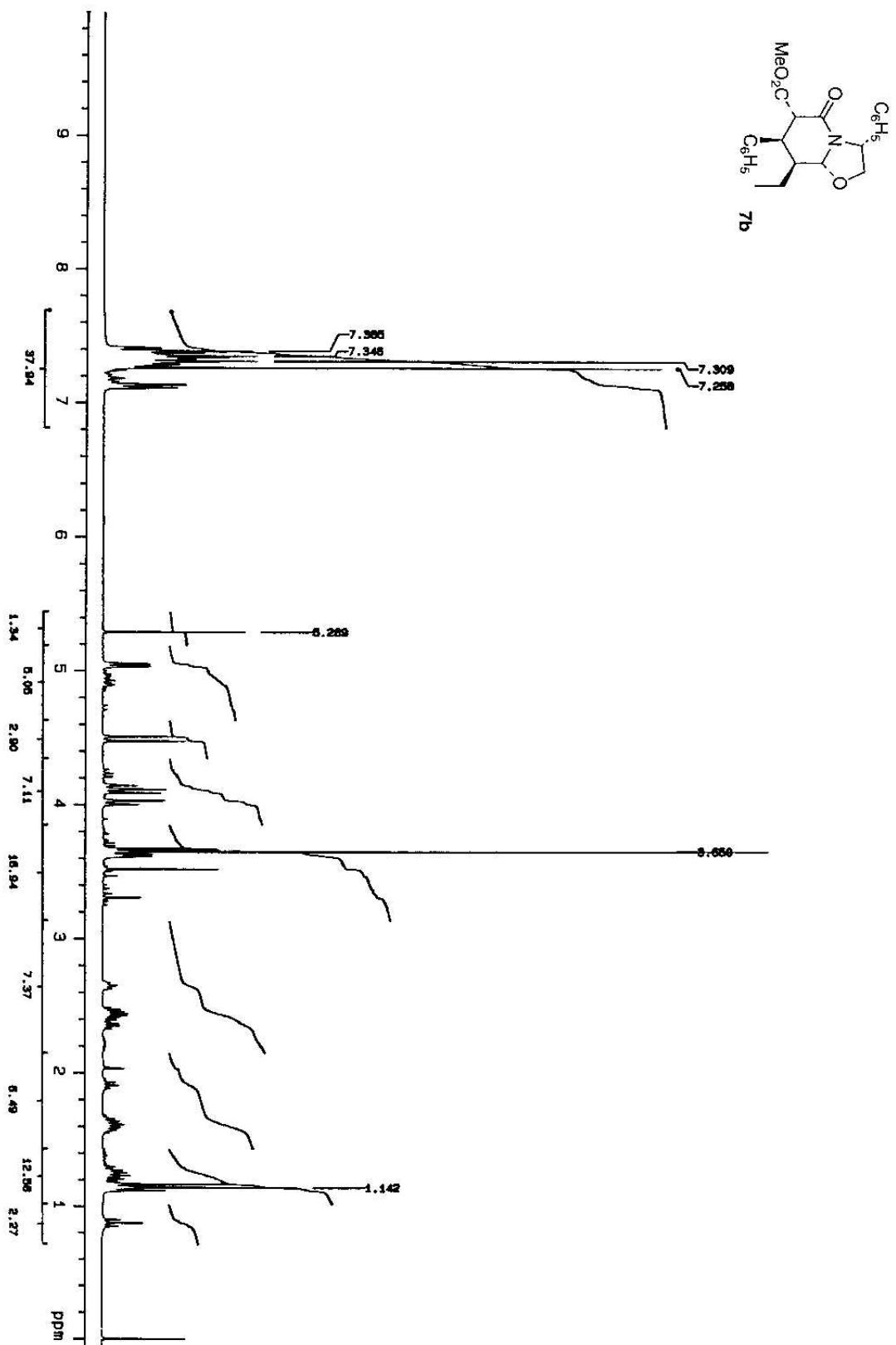


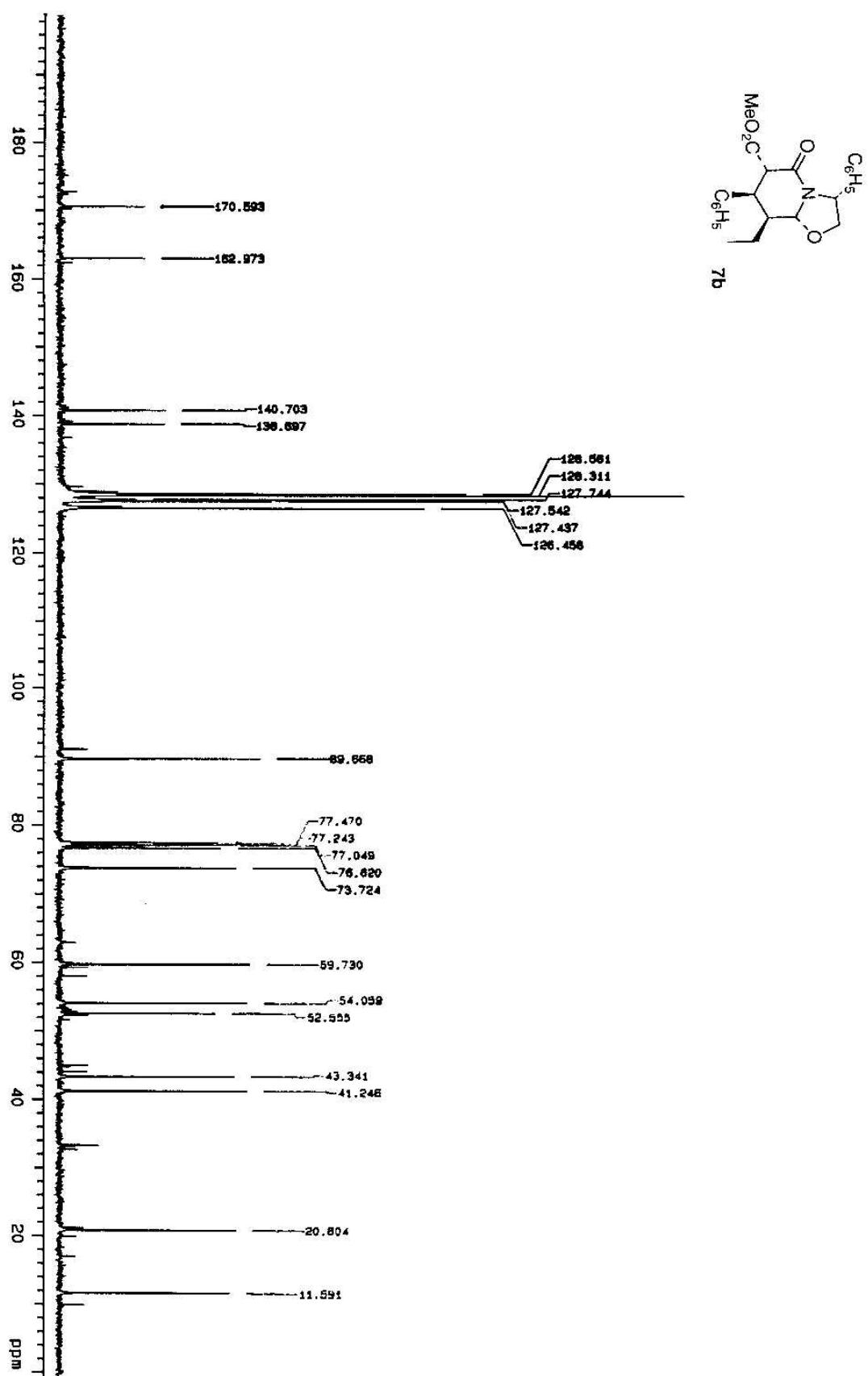


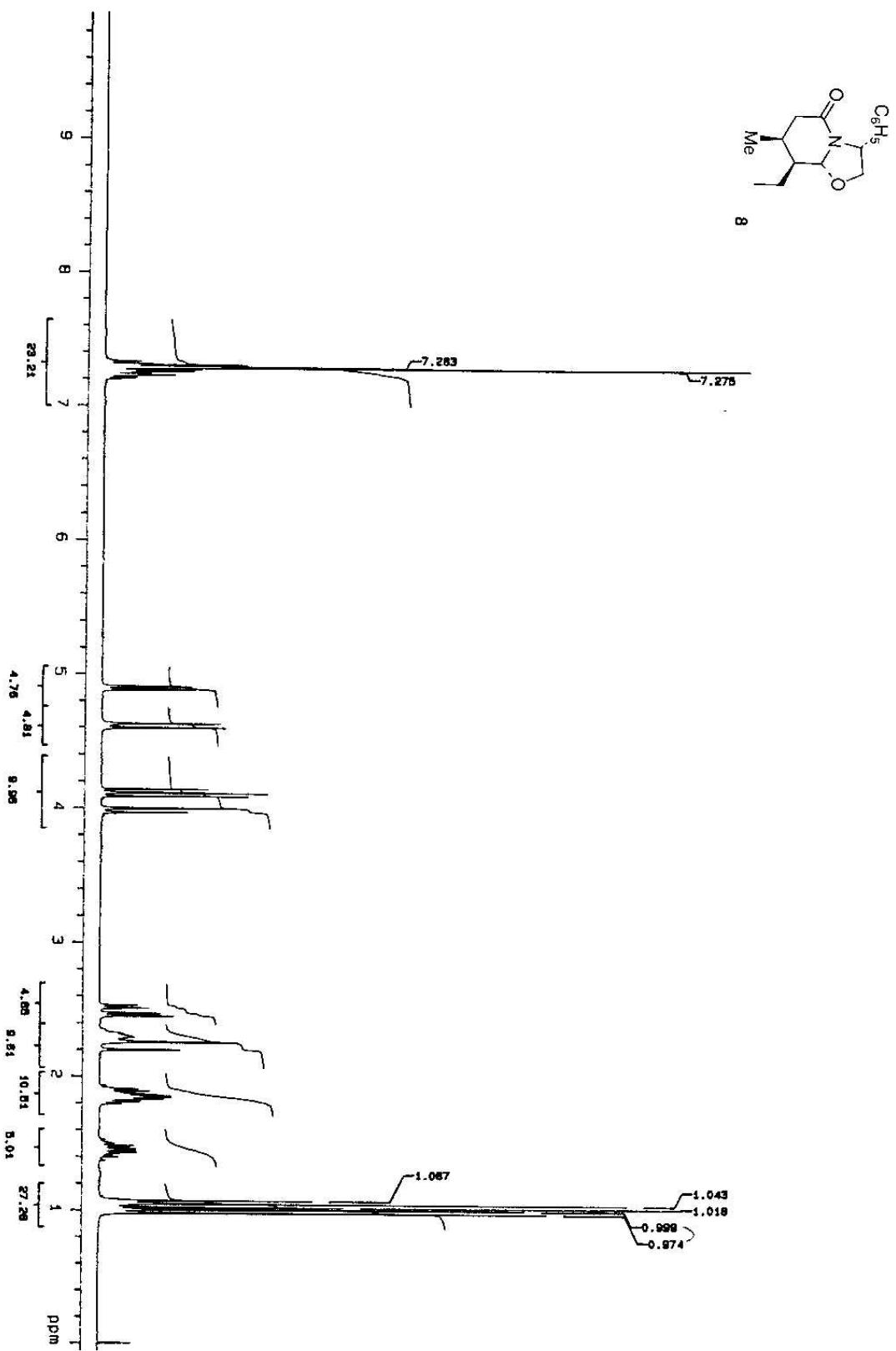


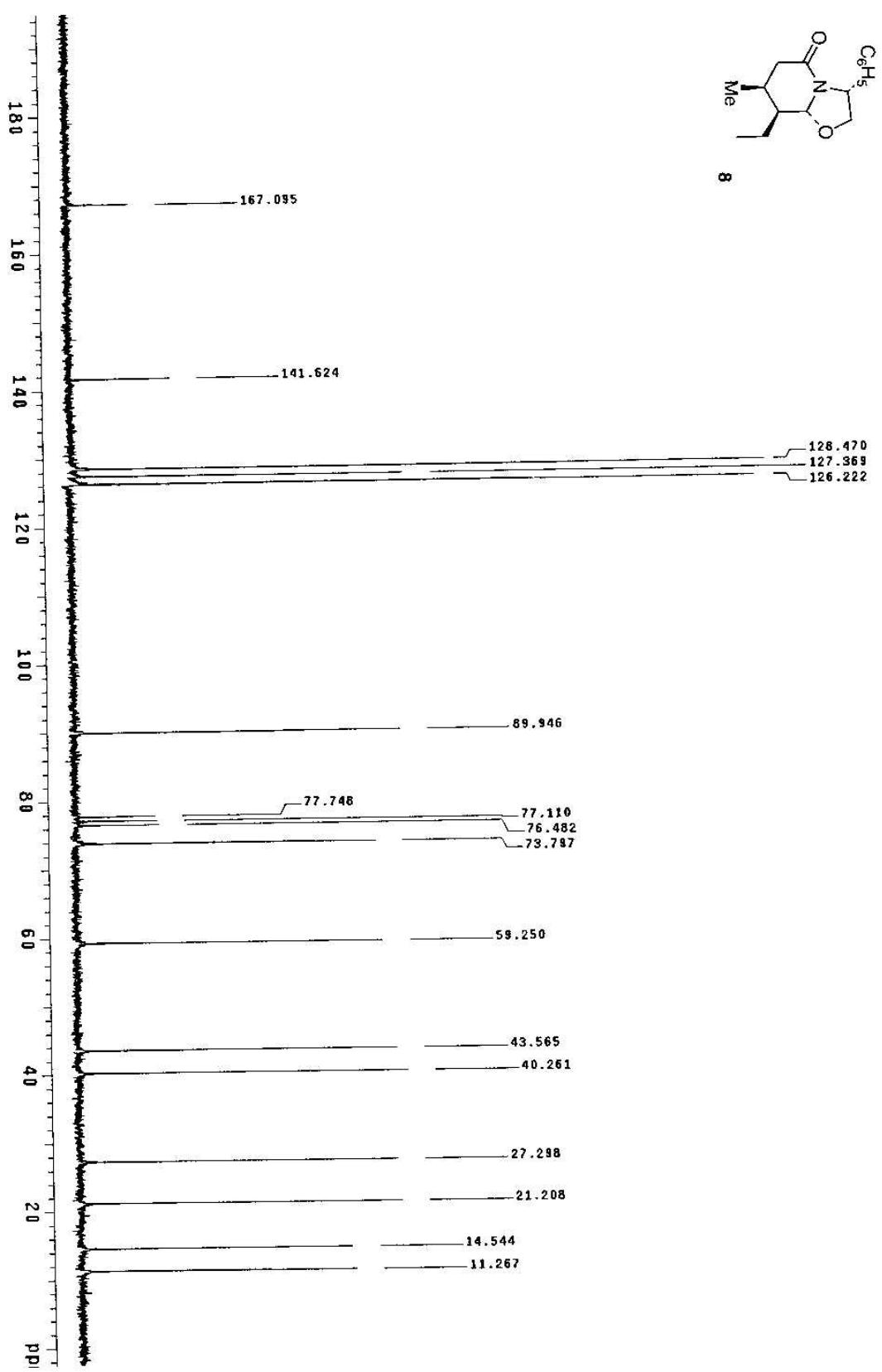


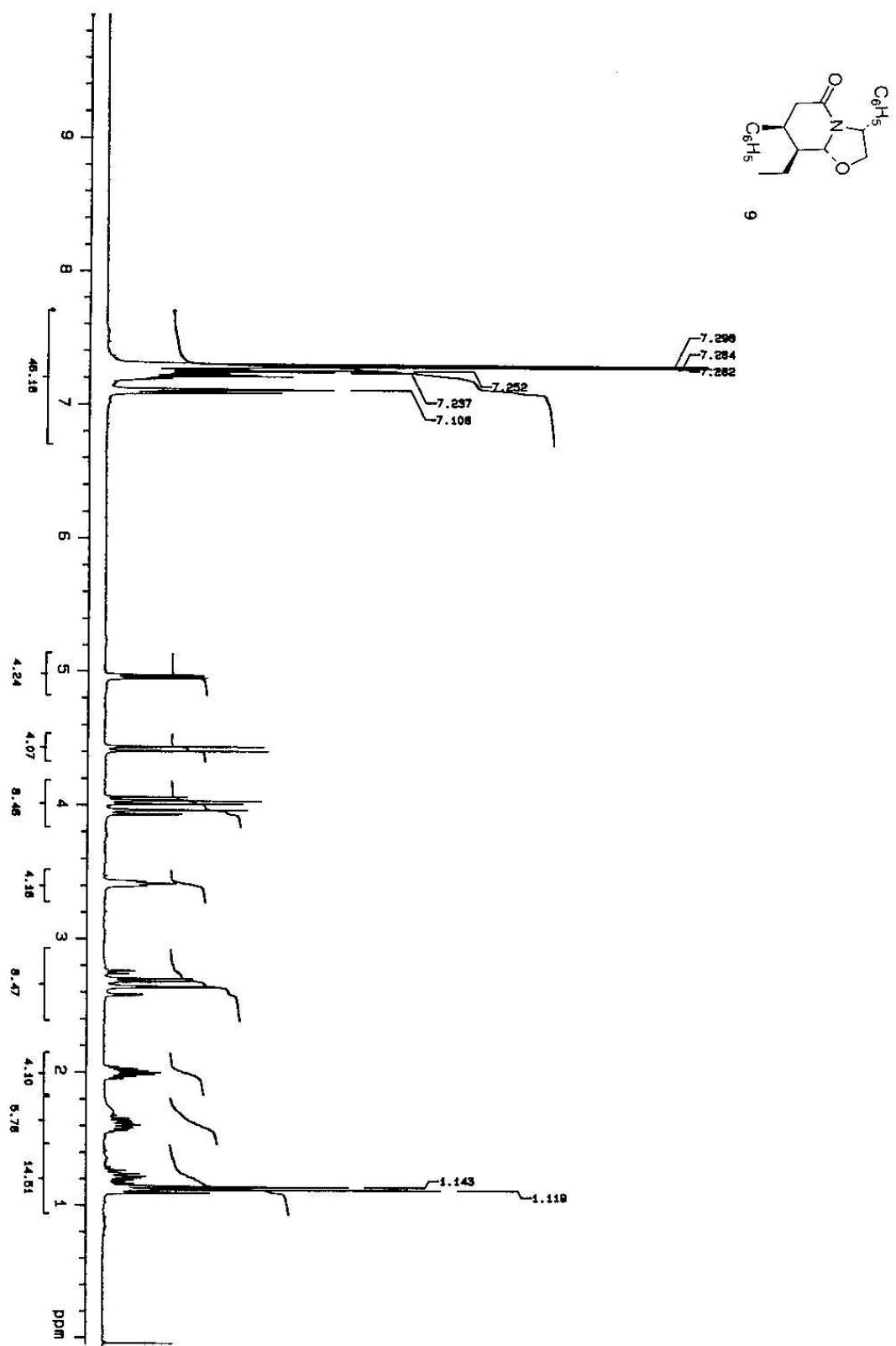


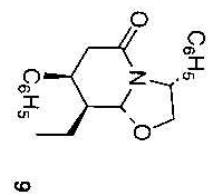
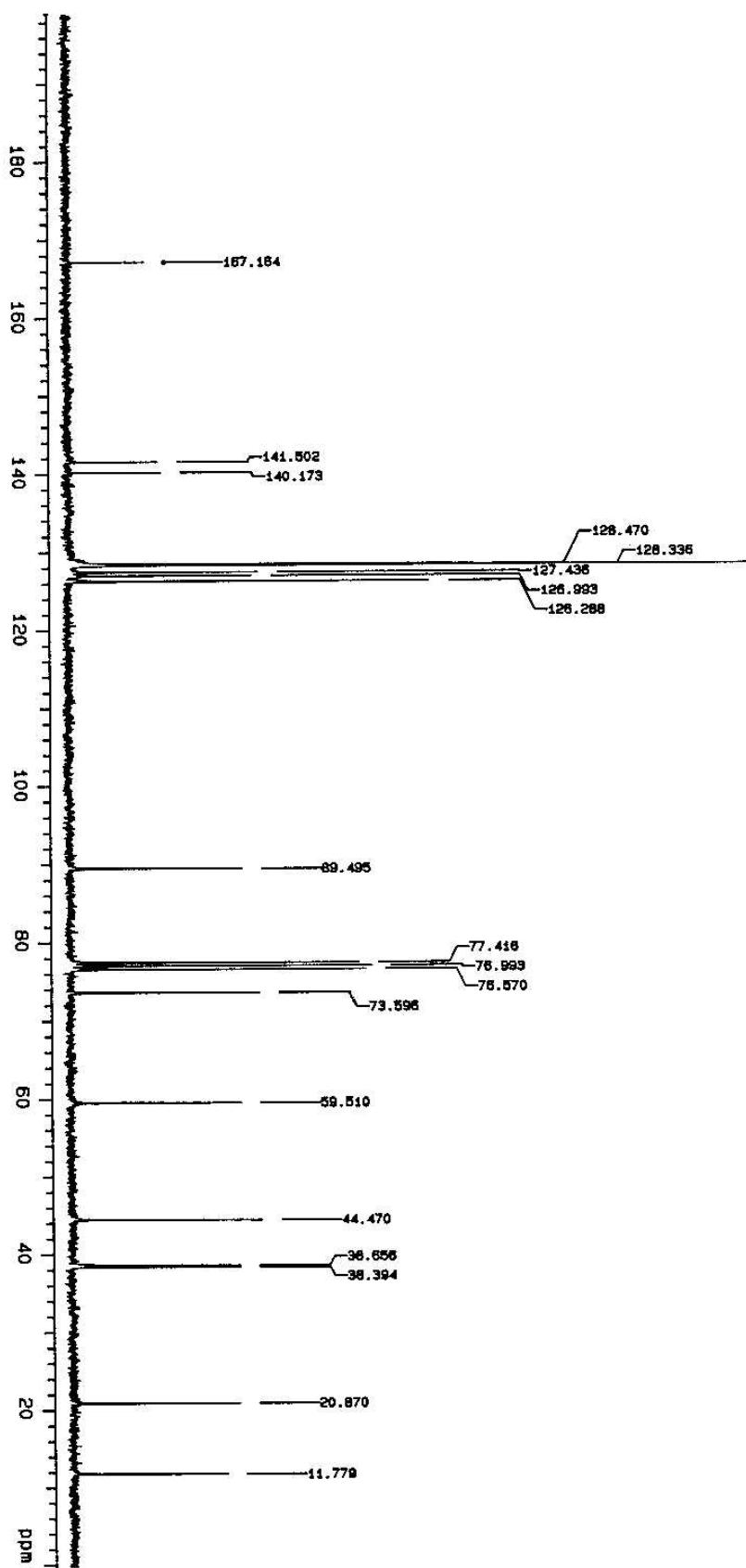


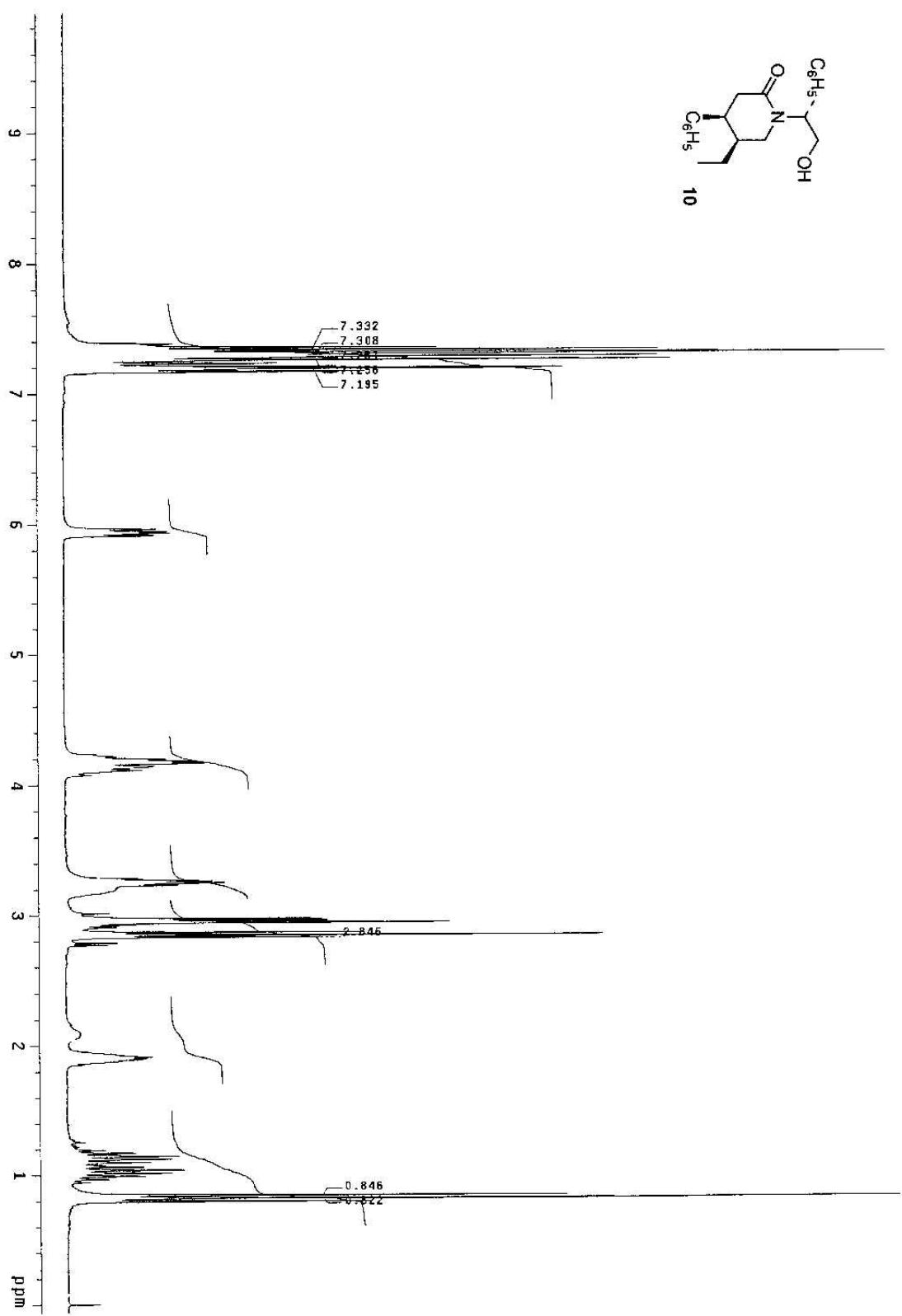
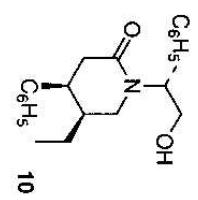


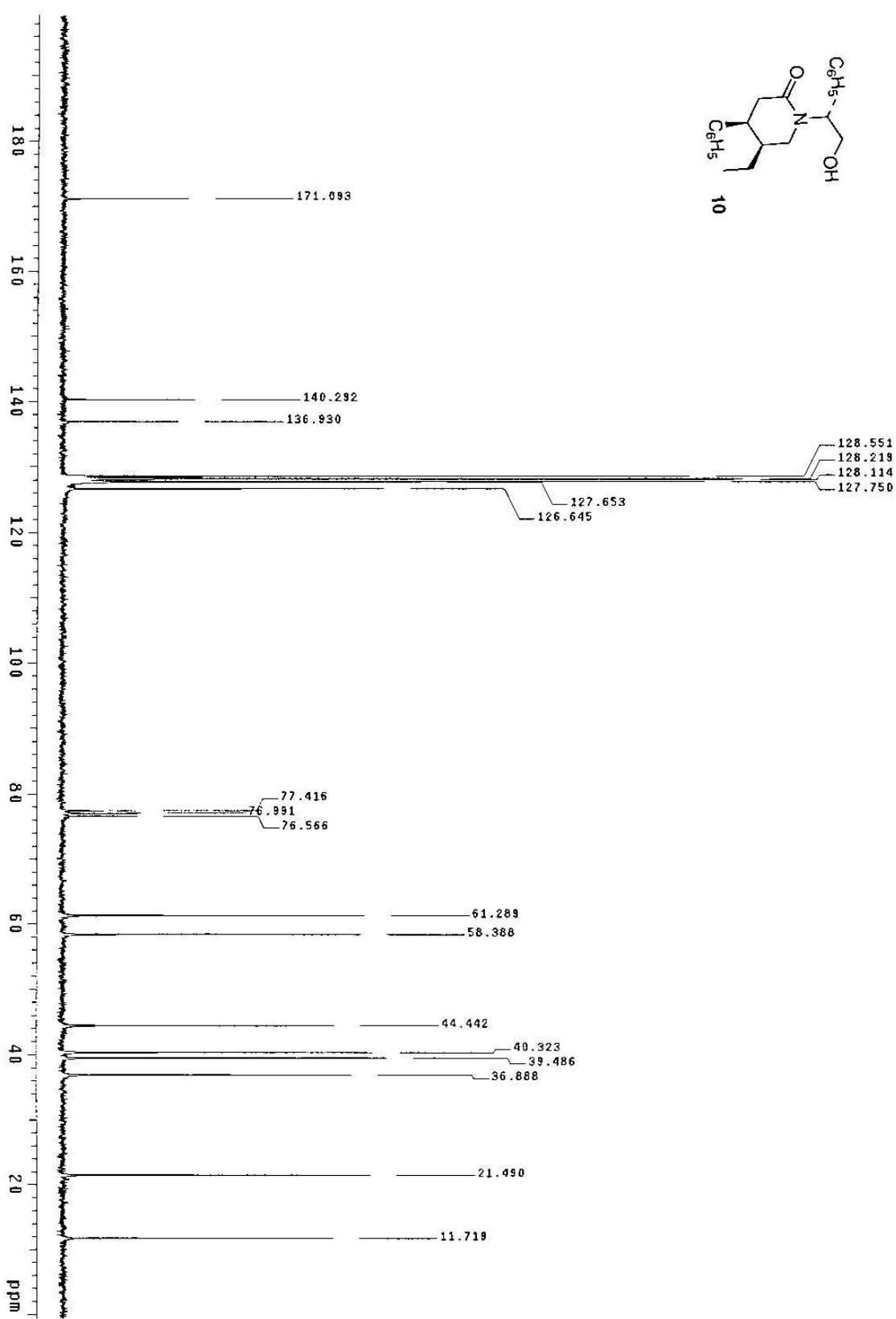


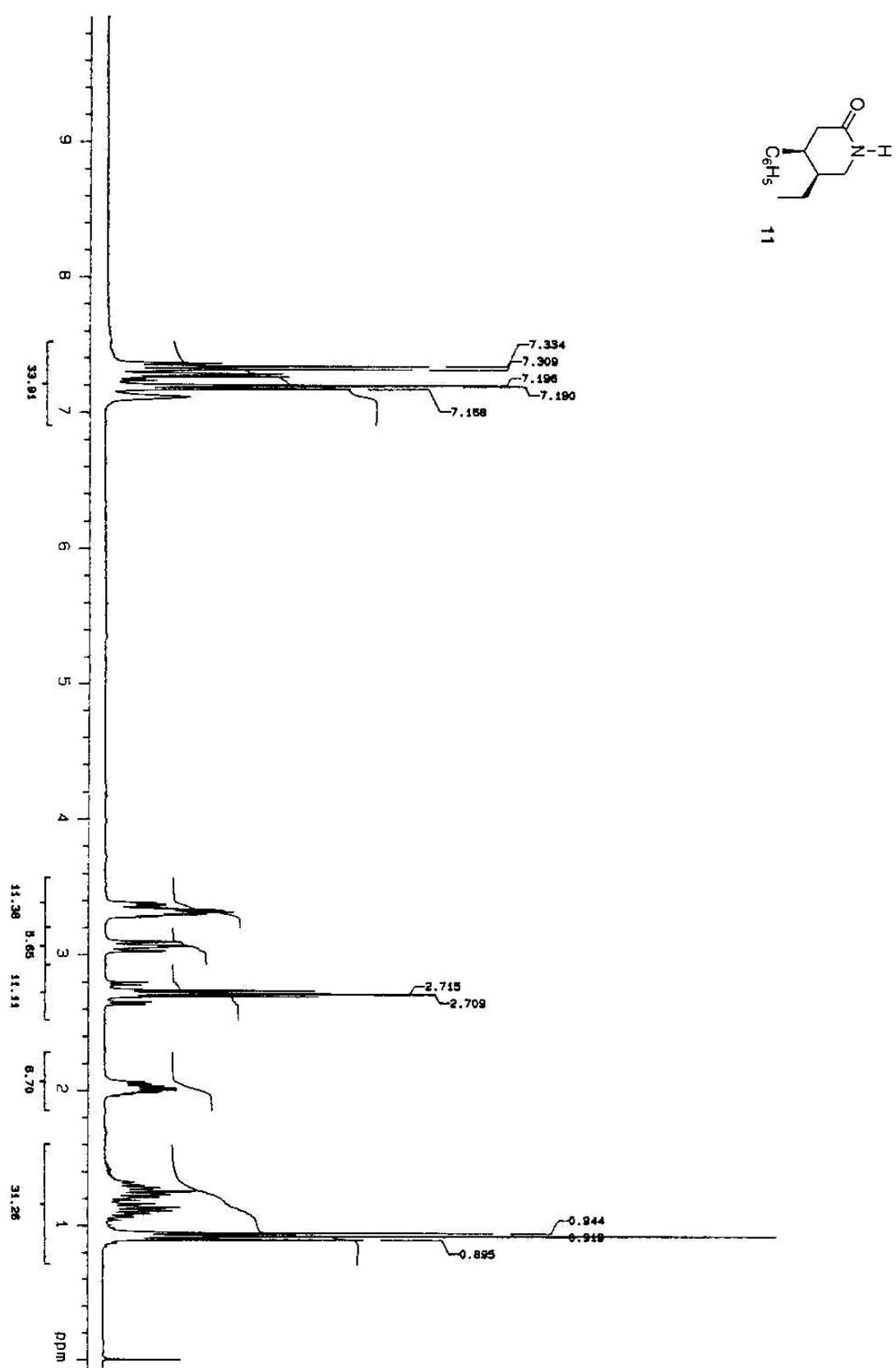


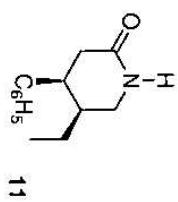
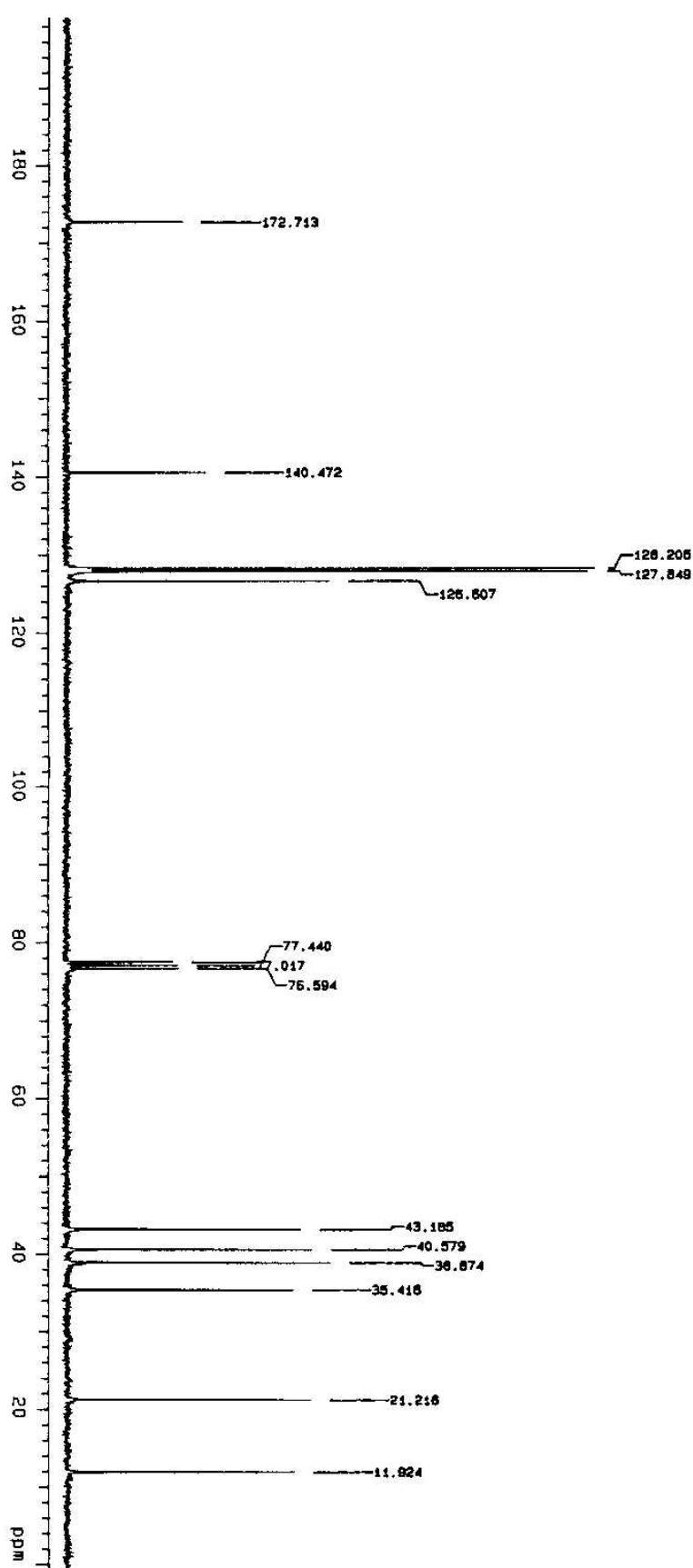


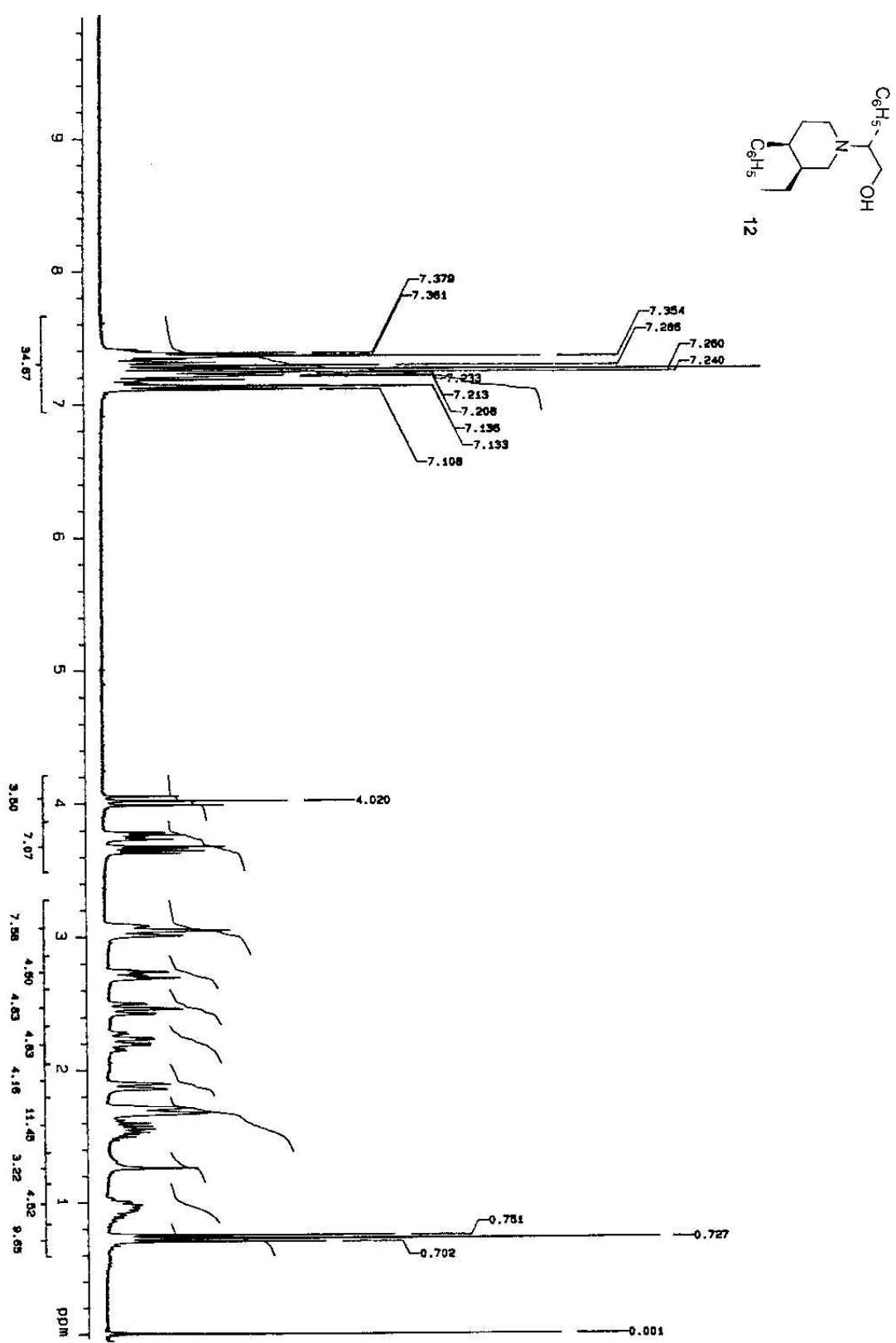


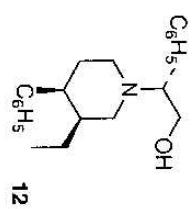
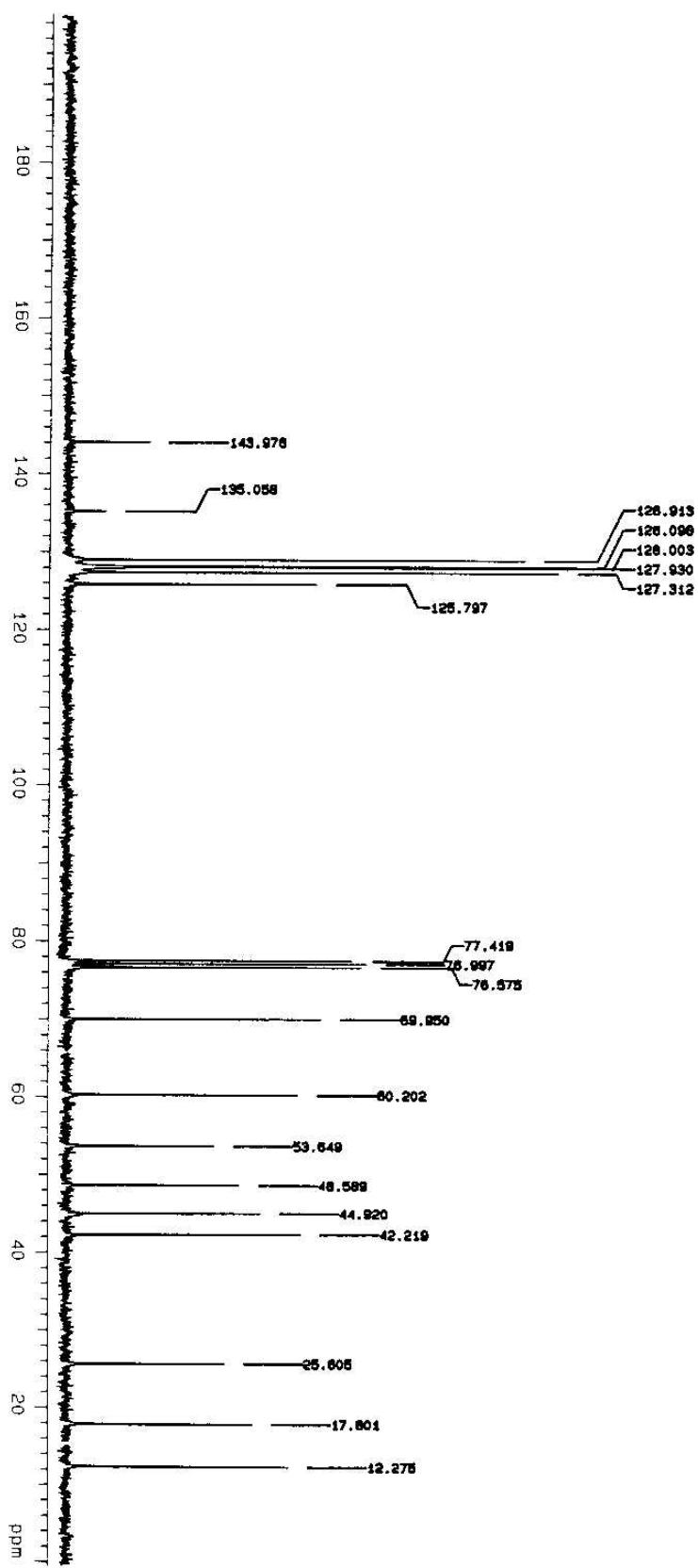


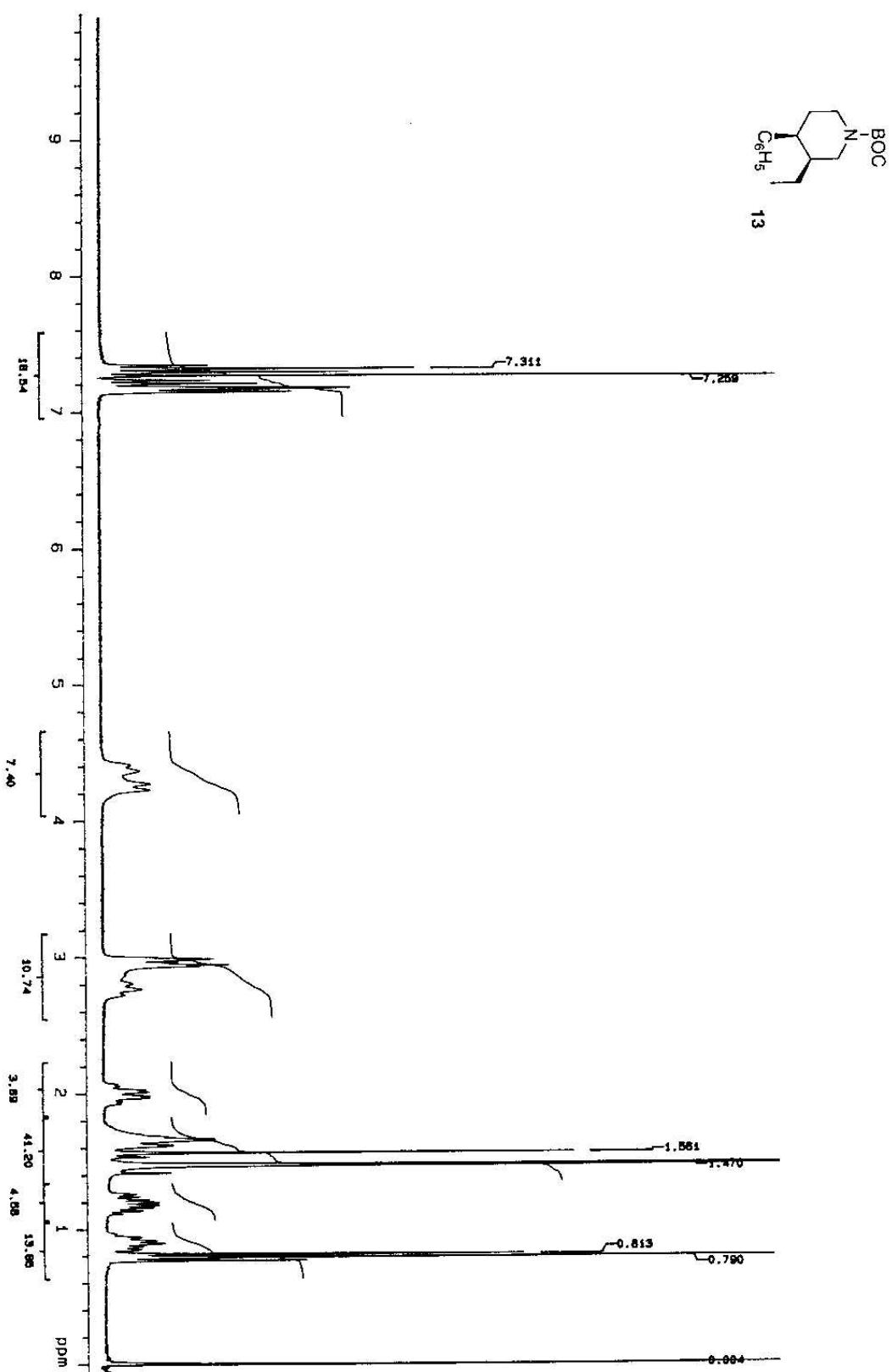


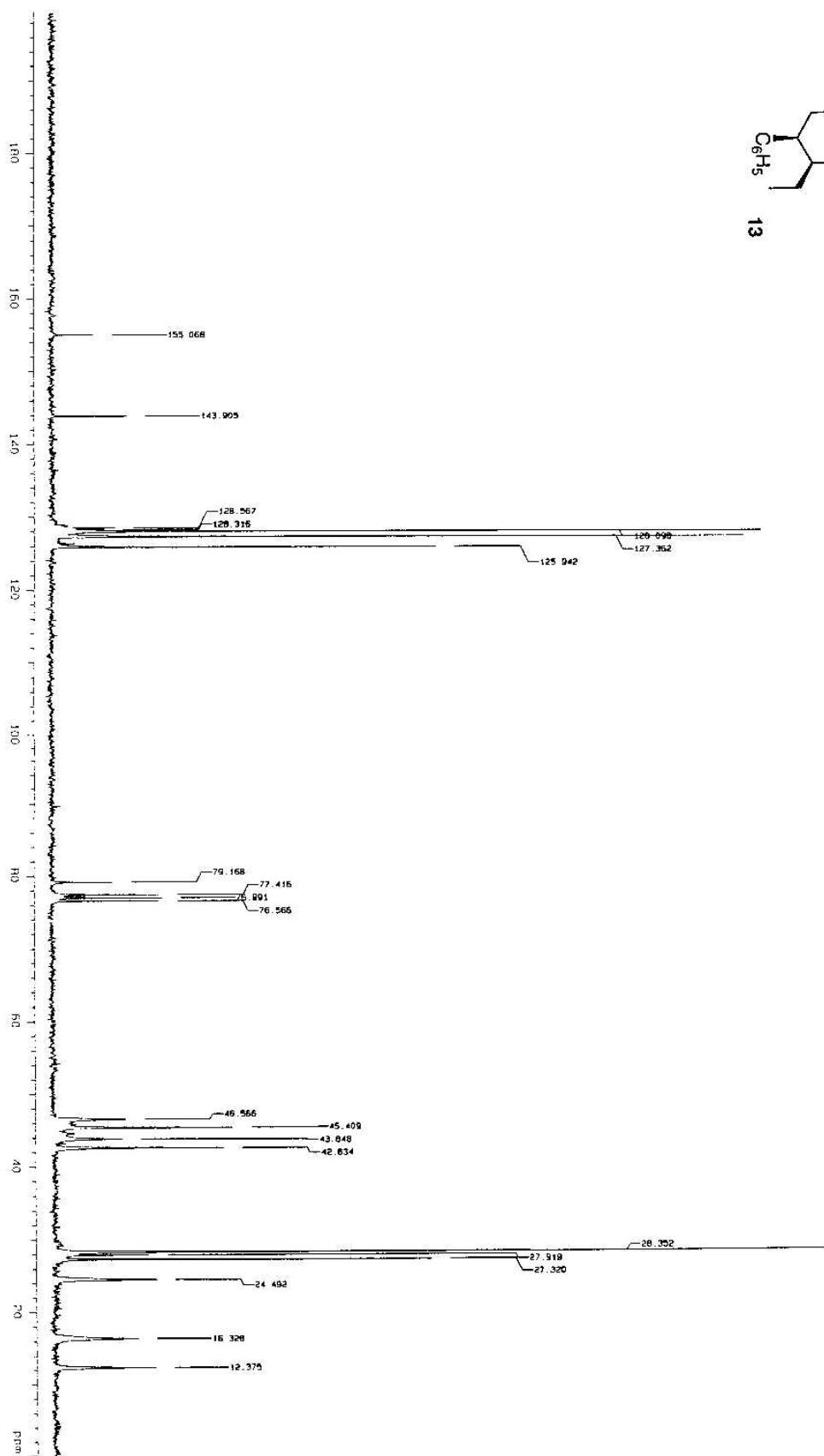


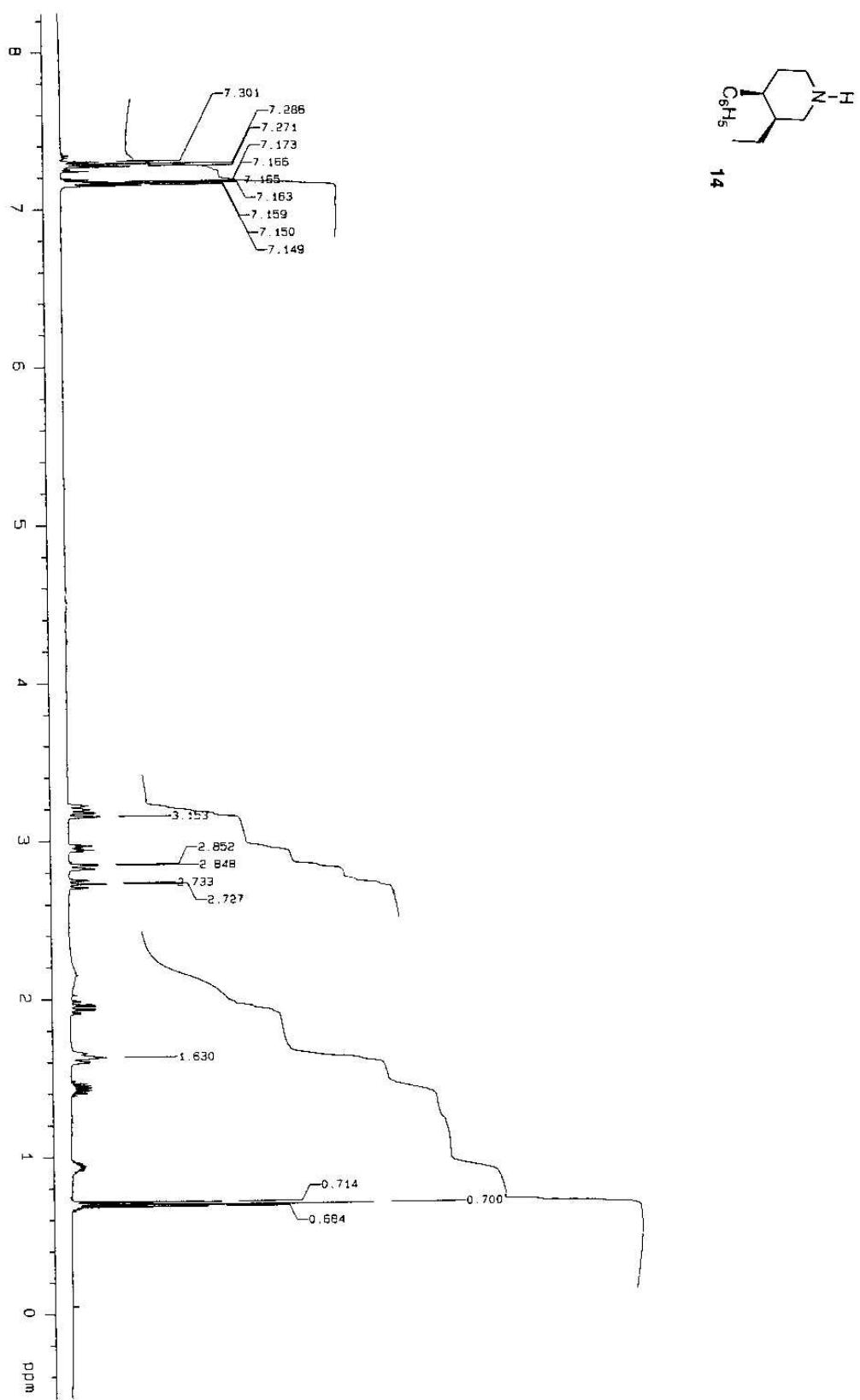


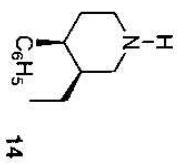
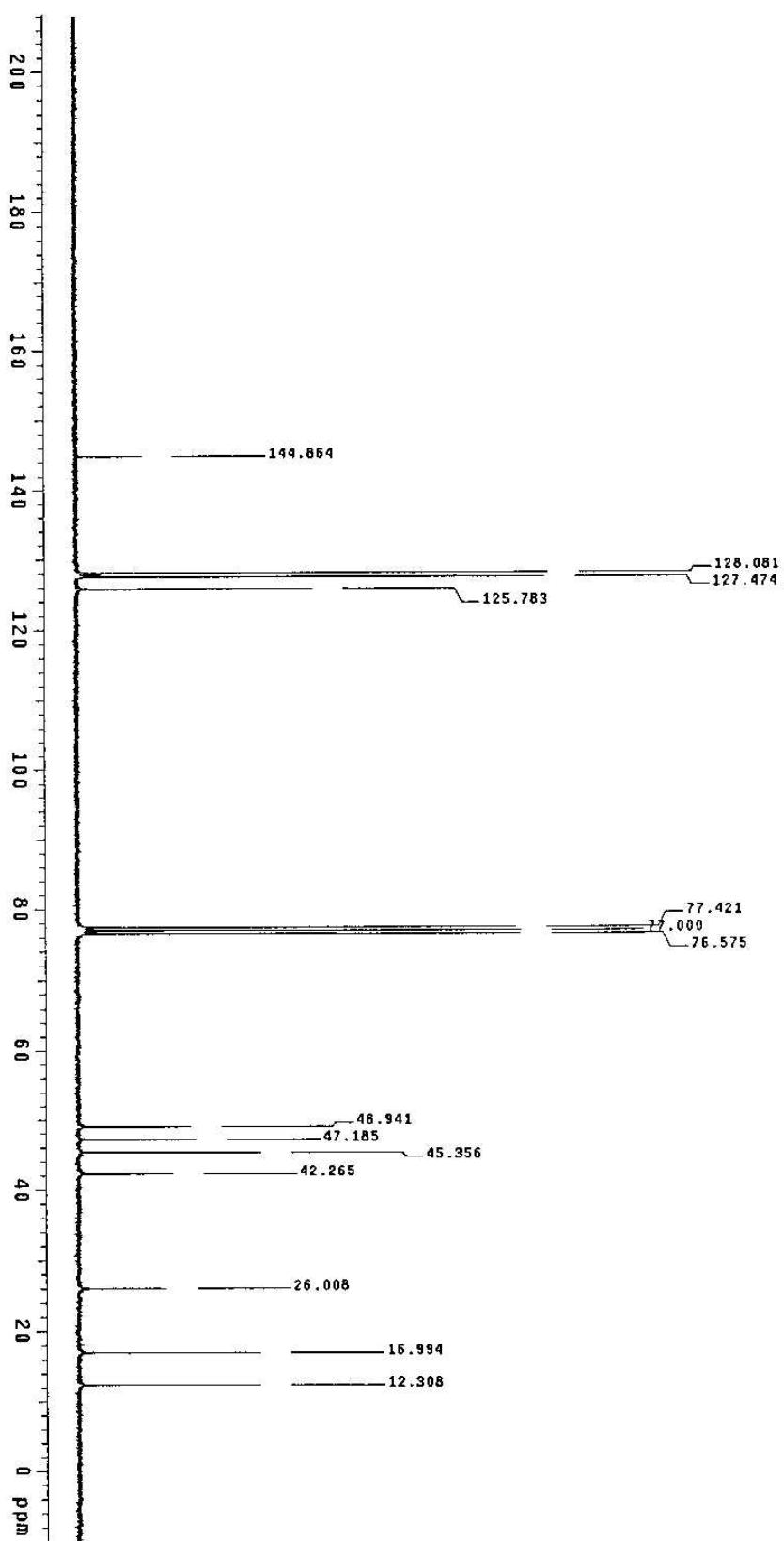


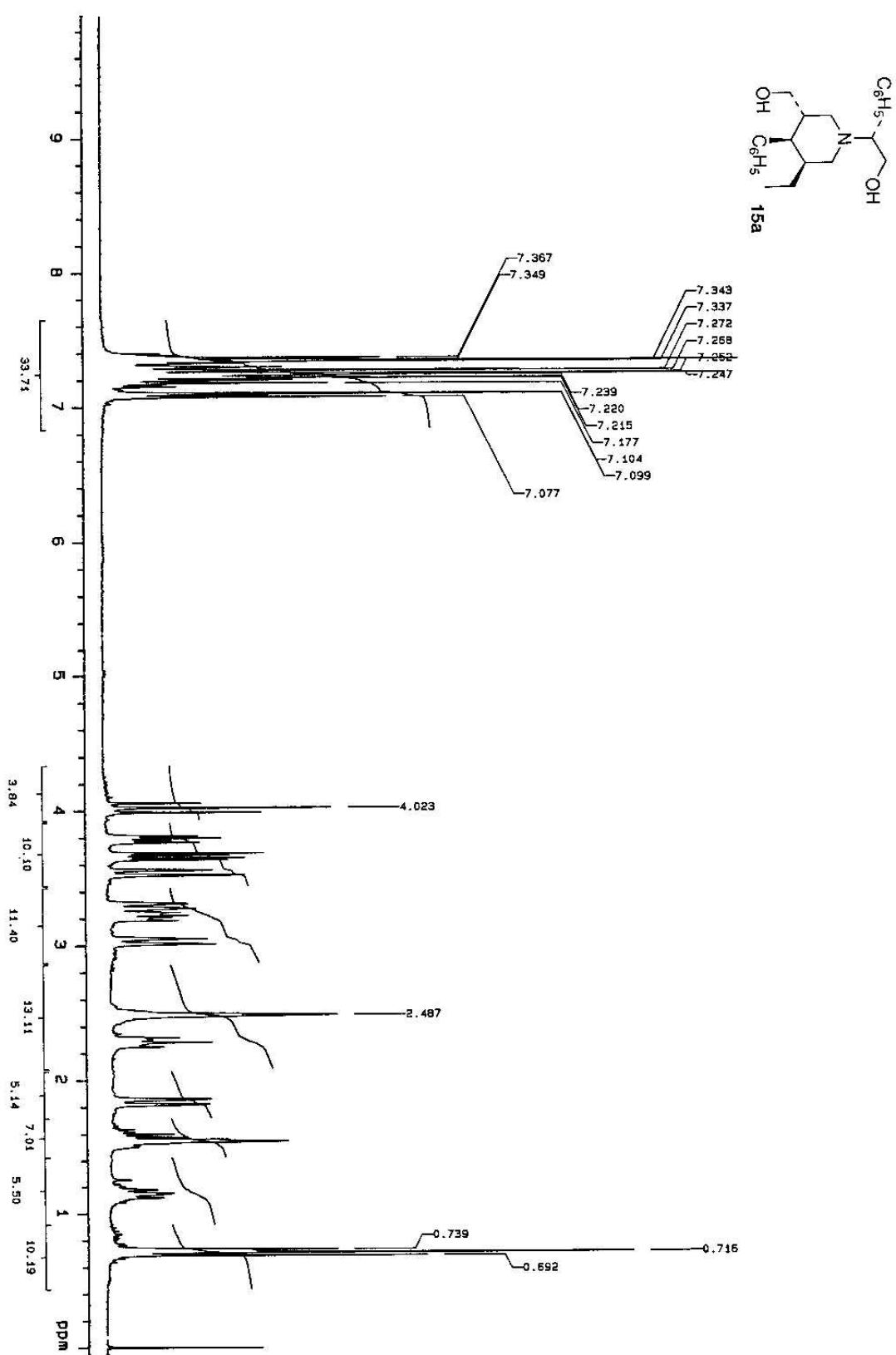


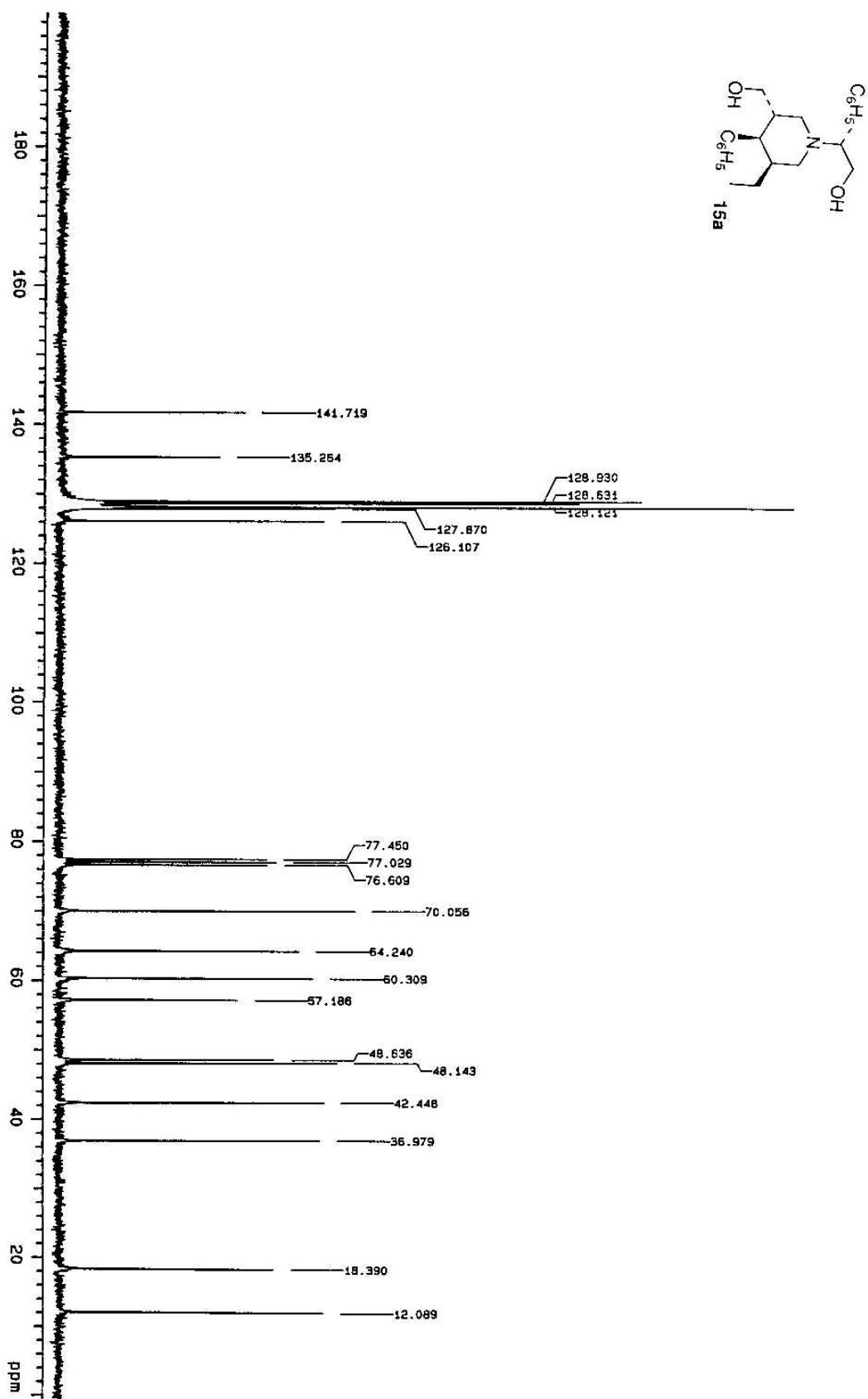


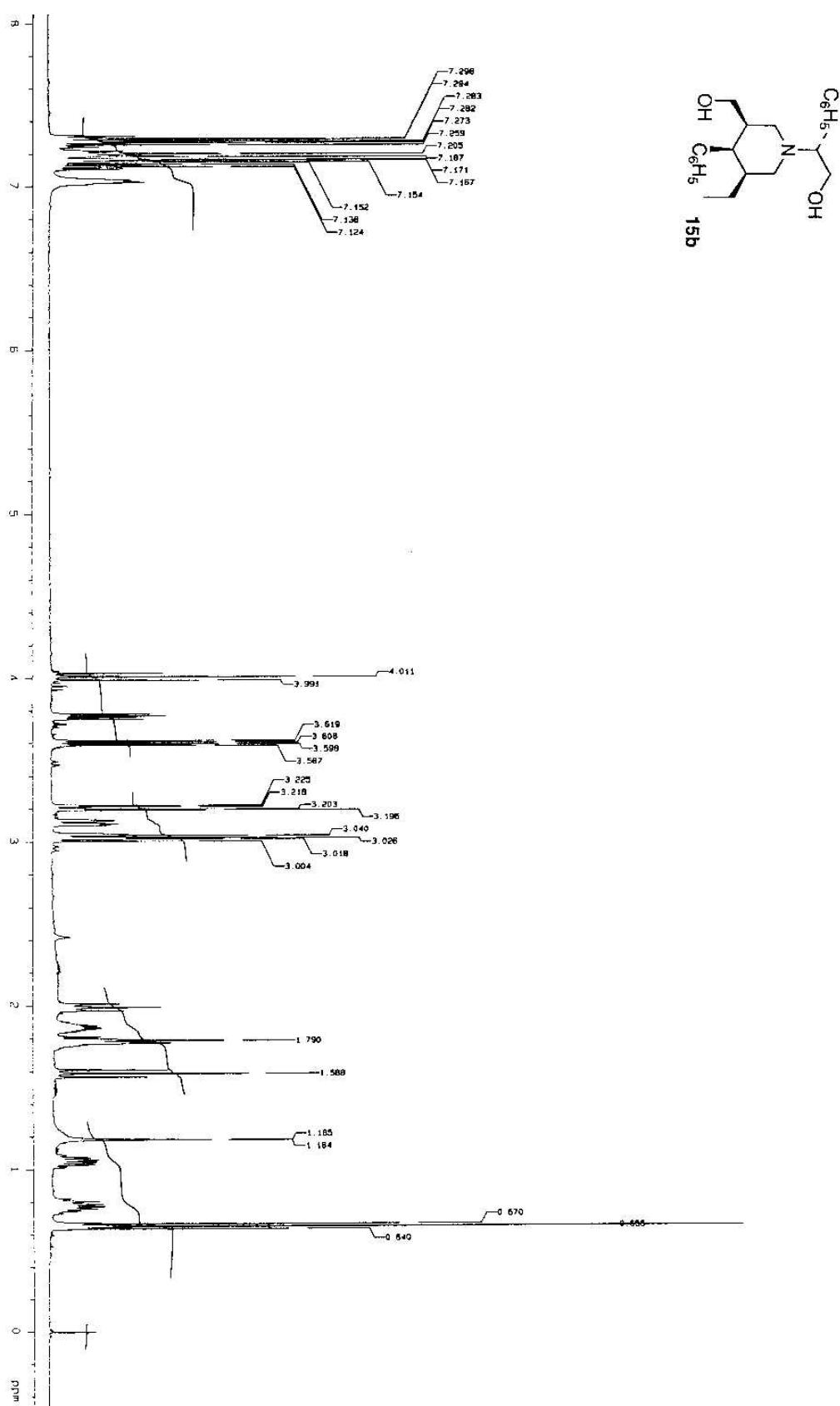


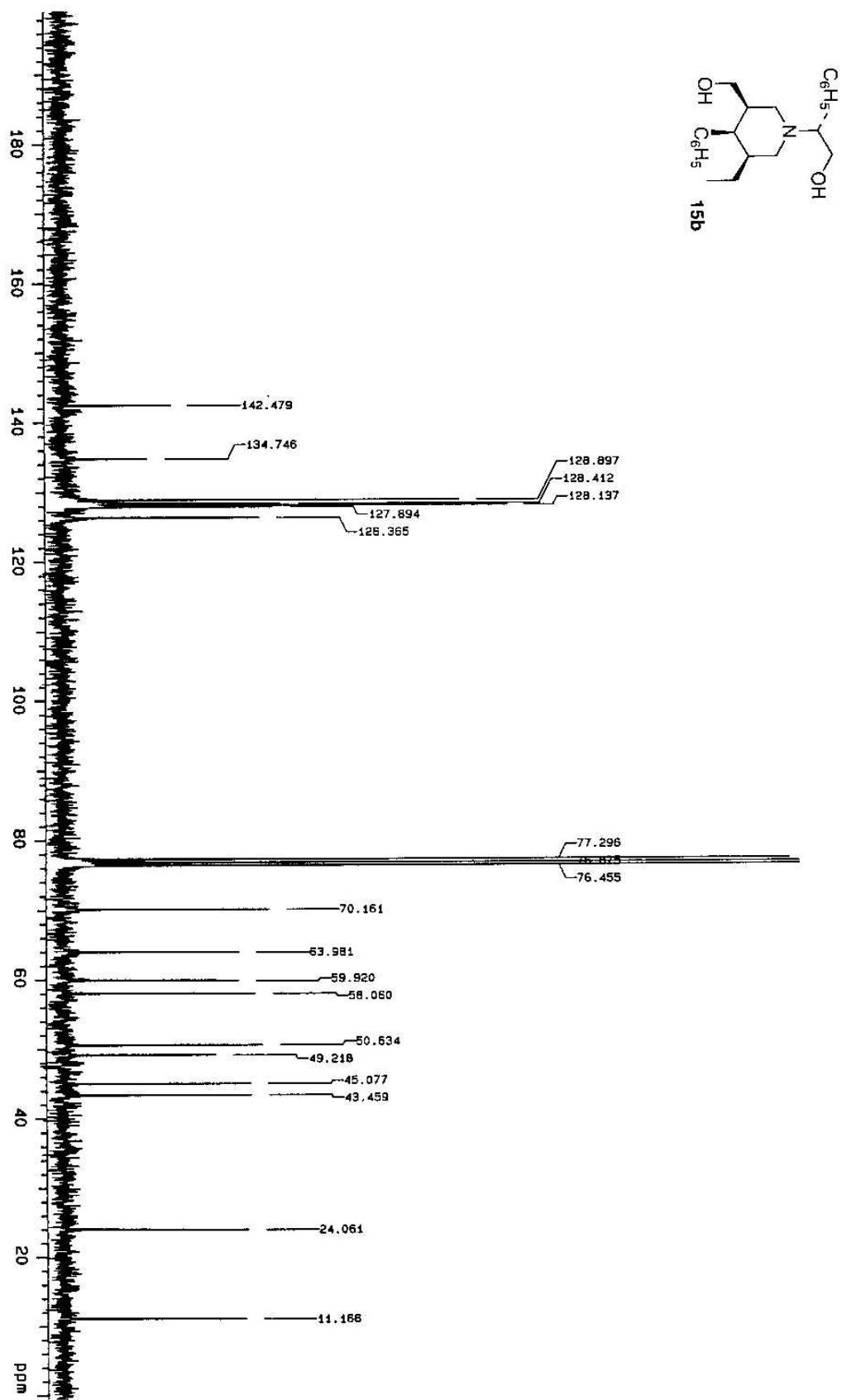


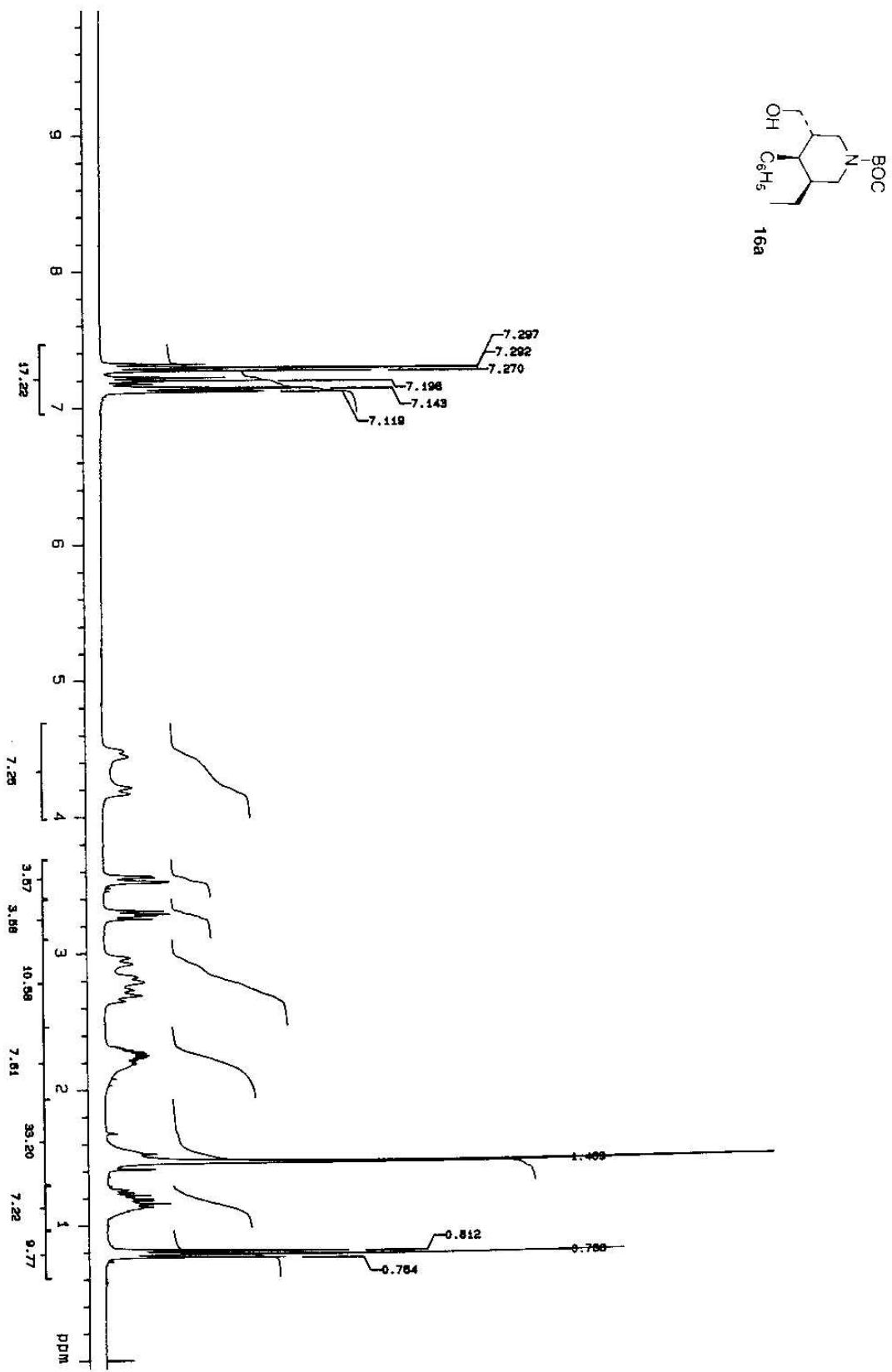


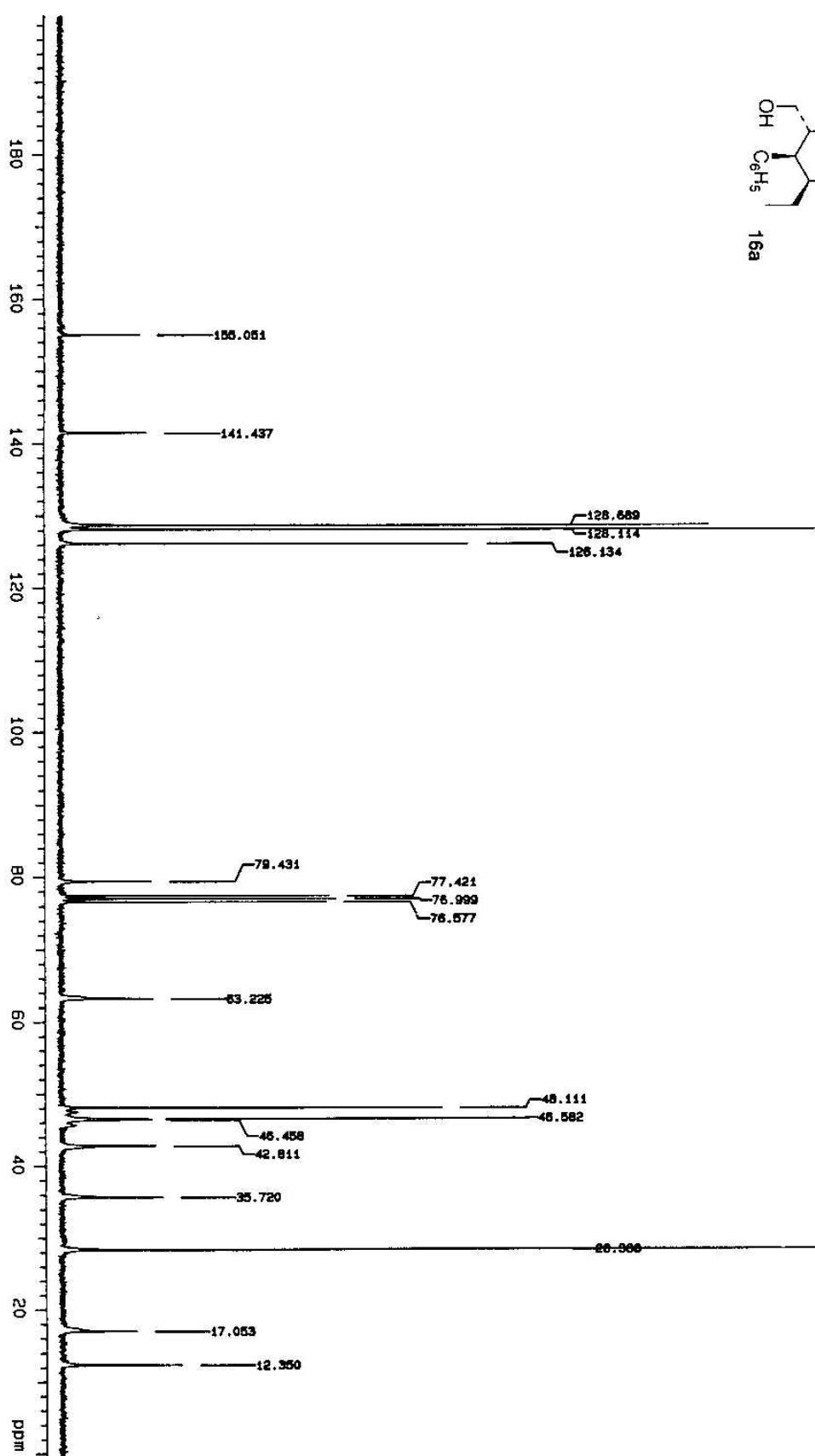


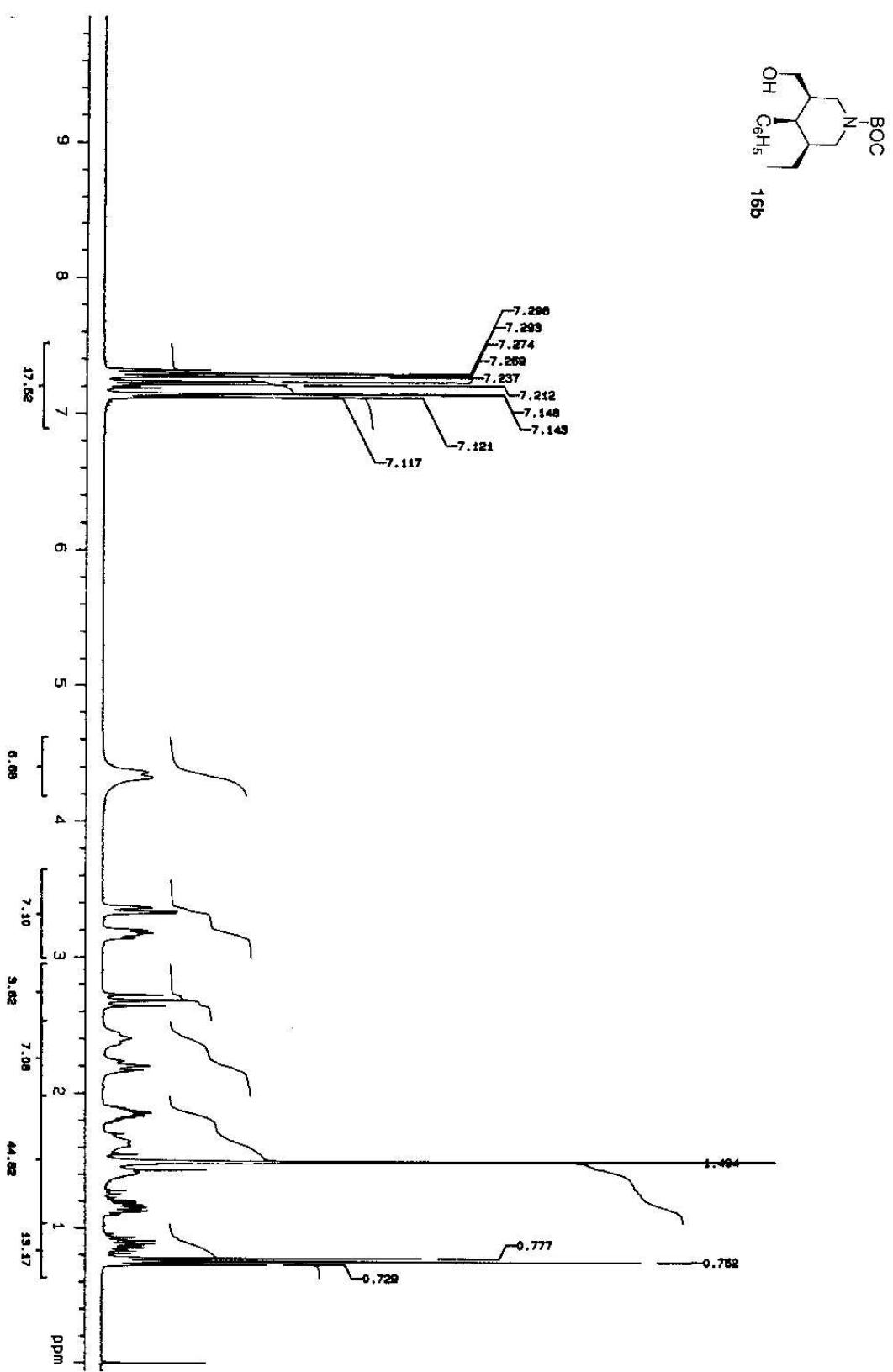


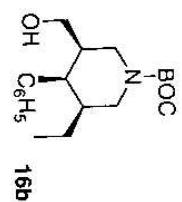
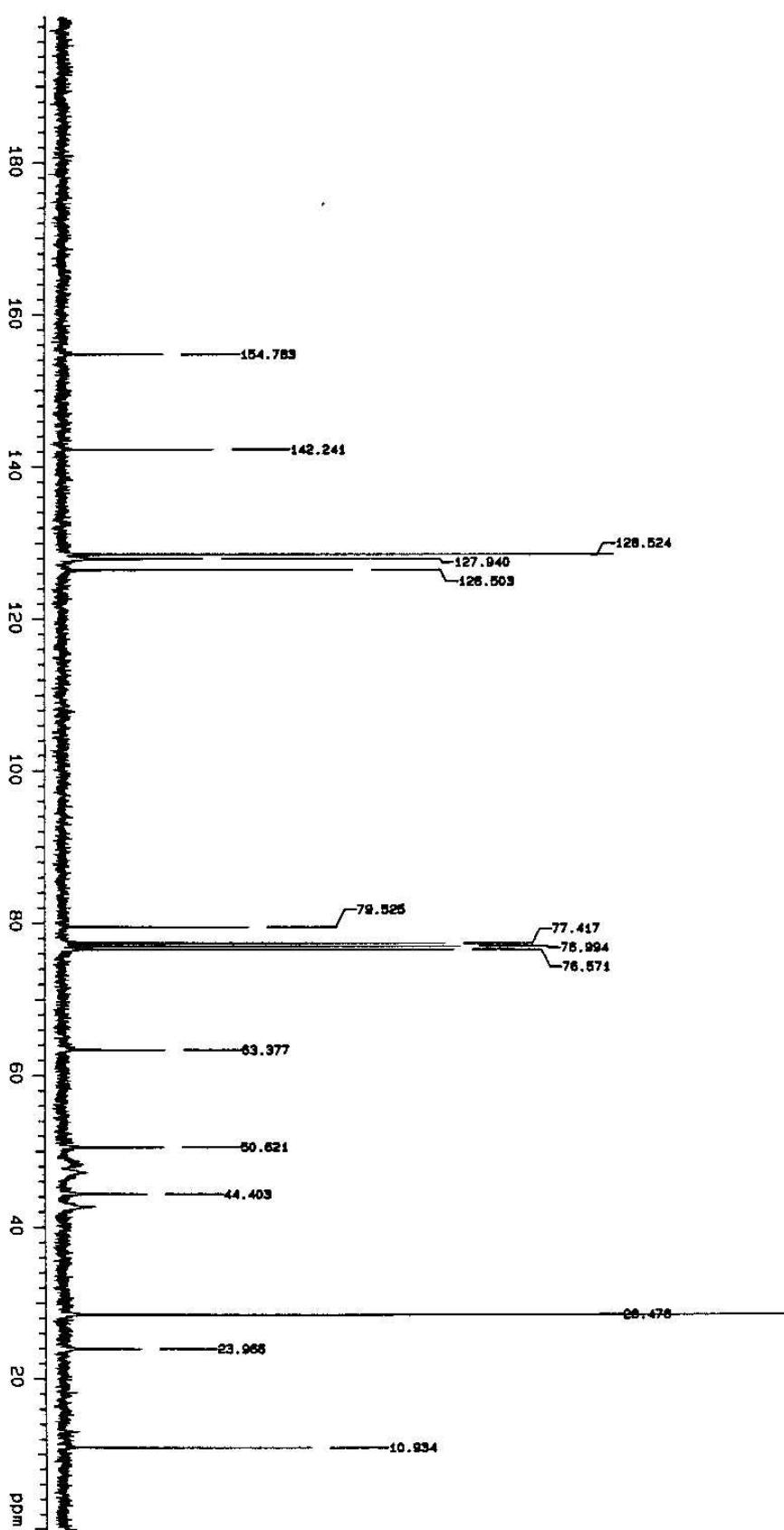


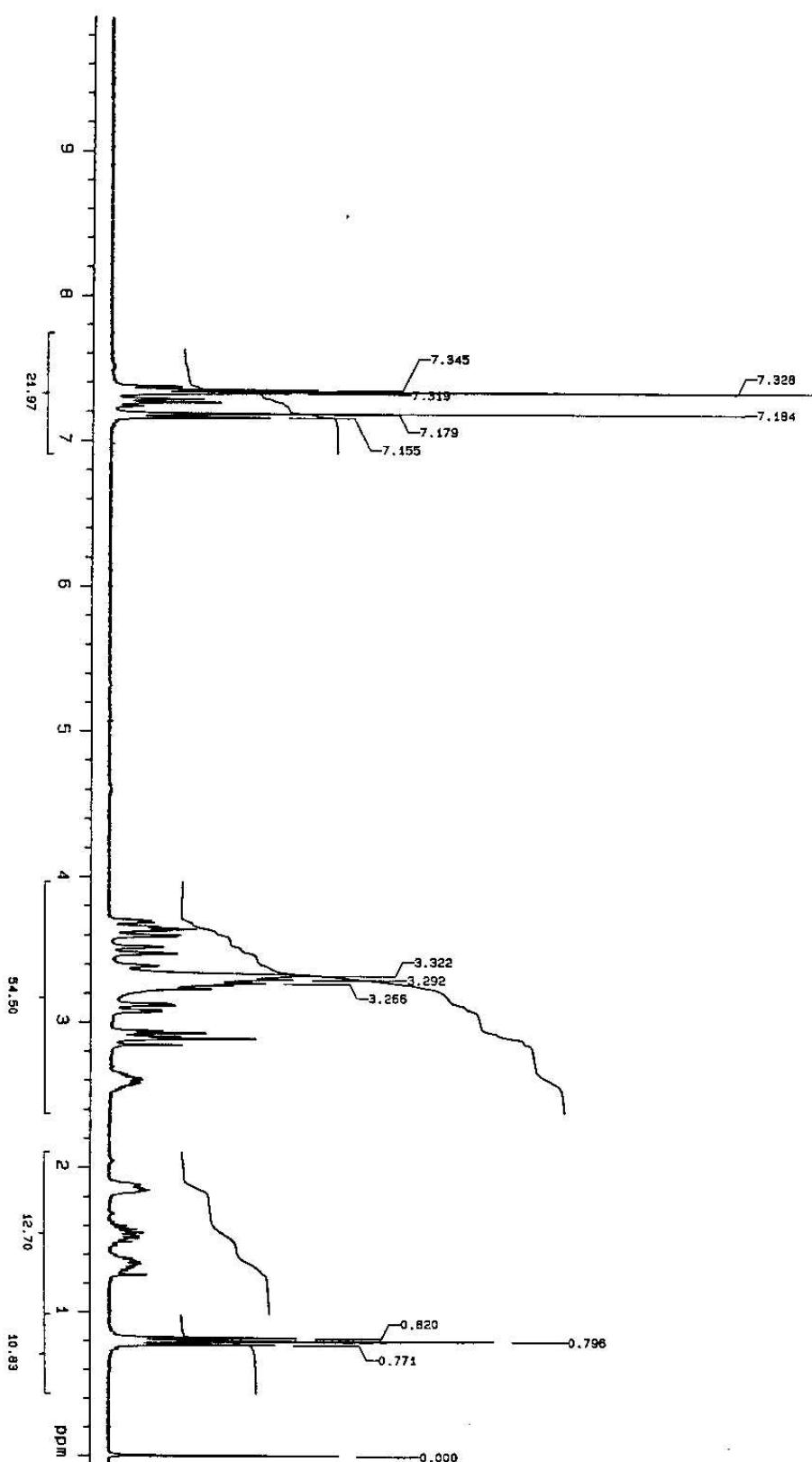
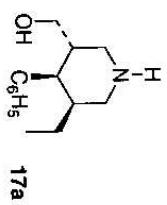


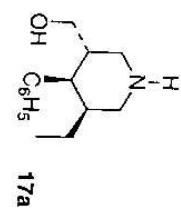
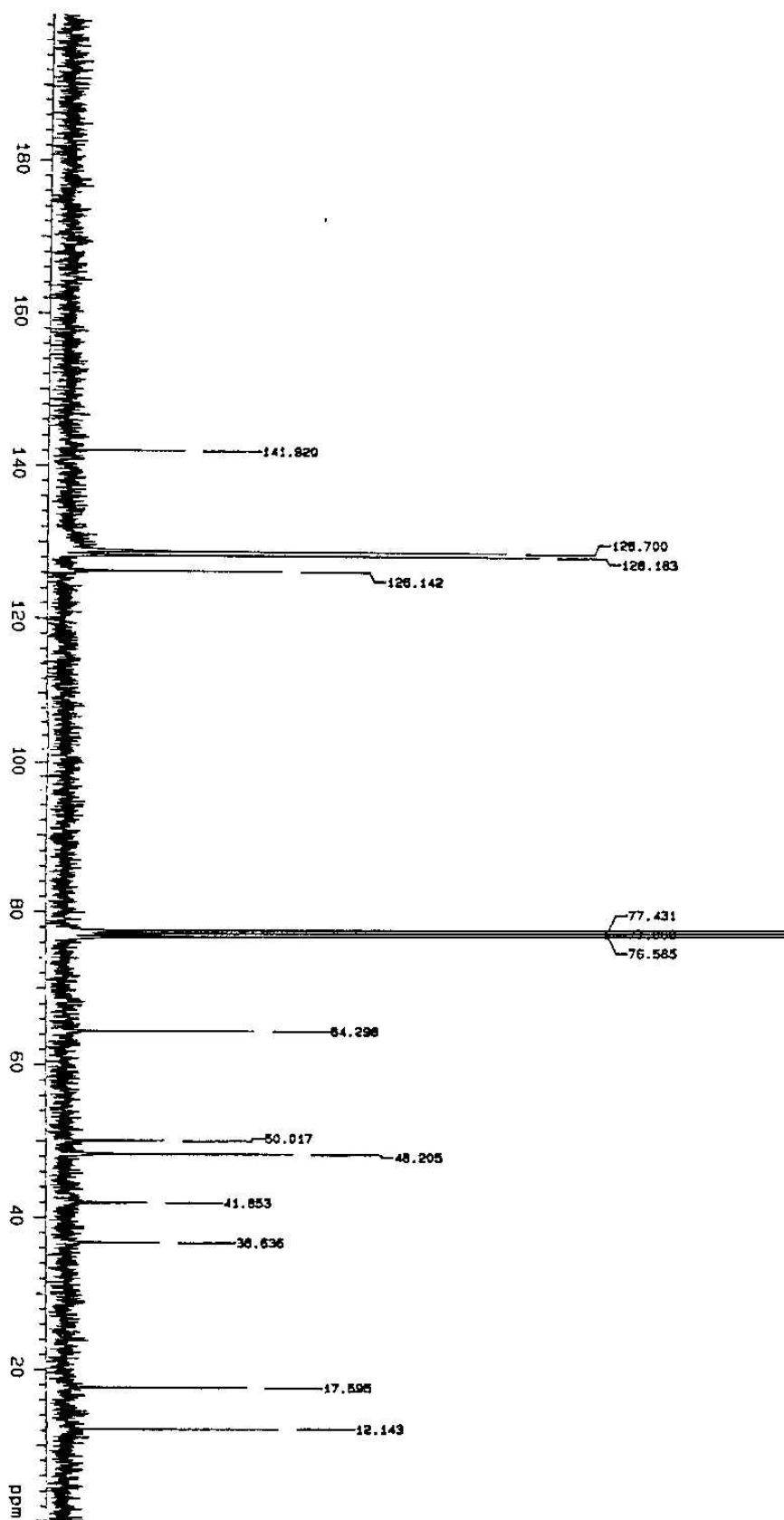












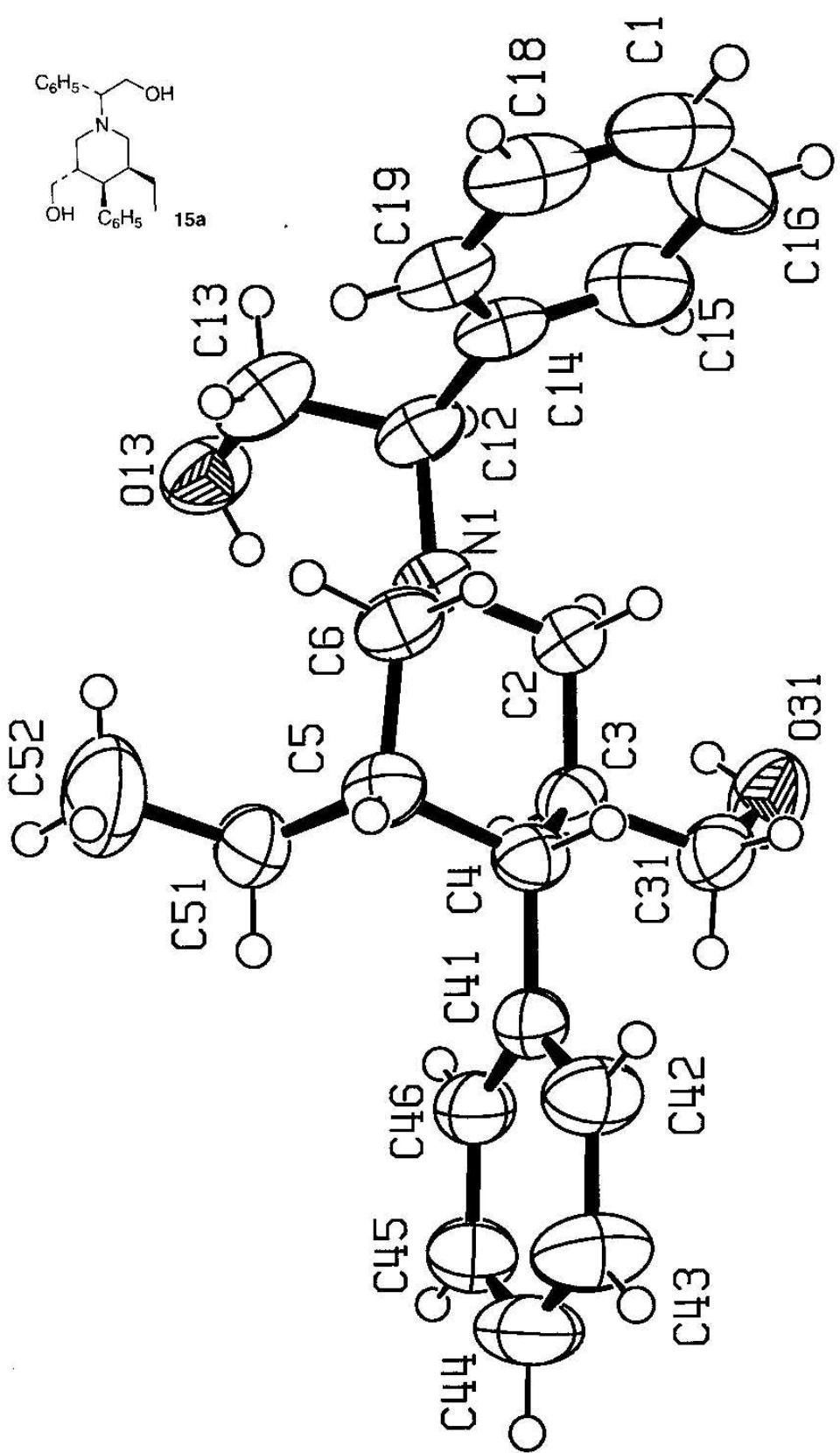


Table 1. Crystal data and structure refinement for 1.

Identification code	jb152
Empirical formula	C22 H29 N O2
Formula weight	339.46
Temperature	293 (2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, C2
Unit cell dimensions	a = 16.570 (4) Å b = 10.150 (4) Å beta = 96.62 (2) deg. c = 11.597 (2) Å
Volume	1937.4 Å ³
Z, Calculated density	4, 1.164 Mg/m ³
Absorption coefficient	0.073 mm ⁻¹
F(000)	736
Crystal size	0.7 x 0.5 x 0.18 mm
Theta range for data collection	2.36 to 28.42 deg.
Limiting indices	0<=h<=22, -13<=k<=0, -15<=l<=15
Reflections collected / unique	2659 / 2159 [R(int) = 0.0204]
Completeness to theta = 28.42	83.9 %
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	2159 / 1 / 250
Goodness-of-fit on F ²	0.958
Final R indices [I>2sigma(I)]	R1 = 0.0470, wR2 = 0.1096
R indices (all data)	R1 = 0.0968, wR2 = 0.1220
Extinction coefficient	0.0047(11)
Largest diff. peak and hole	0.133 and -0.147 e.Å ⁻³

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 1.
 U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
N(1)	-2106(1)	-7939(3)	-3136(2)	49(1)
O(13)	-3051(3)	-9067(6)	-4949(4)	71(1)
O(13A)	-4112(3)	-8932(8)	-4561(4)	101(2)
O(31)	-902(4)	-4148(5)	-3057(5)	80(2)
O(31A)	-489(4)	-4375(8)	-3518(7)	66(2)
C(2)	-1758(2)	-6648(3)	-2826(3)	48(1)
C(3)	-855(2)	-6596(3)	-2962(3)	45(1)
C(31)	-499(2)	-5260(3)	-2577(3)	58(1)
C(4)	-405(2)	-7699(4)	-2248(2)	45(1)
C(41)	510(2)	-7734(3)	-2299(3)	48(1)
C(42)	1028(2)	-7943(4)	-1304(3)	65(1)
C(43)	1862(2)	-8017(5)	-1318(4)	84(1)
C(44)	2188(2)	-7880(5)	-2345(4)	80(1)
C(45)	1685(2)	-7660(4)	-3341(3)	69(1)
C(46)	862(2)	-7572(4)	-3319(3)	58(1)
C(5)	-811(2)	-9057(3)	-2522(3)	51(1)
C(51)	-670(2)	-9625(4)	-3694(3)	63(1)
C(52)	-970(3)	-11019(5)	-3913(5)	115(2)
C(6)	-1710(2)	-8940(4)	-2361(3)	54(1)
C(12)	-3006(2)	-7934(4)	-3217(3)	57(1)
C(13)	-3319(2)	-9071(5)	-3996(3)	80(1)
C(14)	-3356(2)	-7898(4)	-2068(3)	59(1)
C(15)	-3573(3)	-6697(5)	-1602(4)	85(1)
C(16)	-3877(3)	-6645(6)	-539(5)	97(2)
C(17)	-3975(2)	-7769(7)	68(4)	92(1)
C(18)	-3773(2)	-8957(7)	-376(4)	87(1)
C(19)	-3467(2)	-9026(4)	-1427(3)	66(1)

Table 3. Bond lengths [Å] and angles [deg] for 1.

N(1)-C(2)	1.460(4)
N(1)-C(6)	1.461(4)
N(1)-C(12)	1.484(3)
C(2)-C(3)	1.524(4)
C(3)-C(31)	1.525(4)
C(3)-C(4)	1.534(4)
C(31)-O(31)	1.394(6)
C(31)-O(31A)	1.415(8)
C(4)-C(41)	1.524(4)
C(4)-C(5)	1.550(5)
C(41)-C(42)	1.372(4)
C(41)-C(46)	1.388(4)
C(42)-C(43)	1.386(4)
C(43)-C(44)	1.371(5)
C(44)-C(45)	1.362(5)
C(45)-C(46)	1.370(4)
C(5)-C(51)	1.519(5)
C(5)-C(6)	1.527(4)
C(51)-C(52)	1.511(6)
C(12)-C(14)	1.515(4)
C(12)-C(13)	1.520(5)
C(13)-O(13)	1.237(5)
C(13)-O(13A)	1.406(6)
C(14)-C(19)	1.388(5)
C(14)-C(15)	1.397(6)
C(15)-C(16)	1.385(6)
C(16)-C(17)	1.360(7)
C(17)-C(18)	1.369(7)
C(18)-C(19)	1.375(6)
C(2)-N(1)-C(6)	109.7(2)
C(2)-N(1)-C(12)	112.2(3)
C(6)-N(1)-C(12)	114.8(2)
N(1)-C(2)-C(3)	111.7(2)
C(2)-C(3)-C(31)	110.6(3)
C(2)-C(3)-C(4)	109.9(2)
C(31)-C(3)-C(4)	110.1(3)
O(31)-C(31)-O(31A)	39.2(3)
O(31)-C(31)-C(3)	116.9(3)
O(31A)-C(31)-C(3)	112.2(4)
C(41)-C(4)-C(3)	114.8(3)
C(41)-C(4)-C(5)	112.4(3)
C(3)-C(4)-C(5)	111.5(2)
C(42)-C(41)-C(46)	116.8(3)
C(42)-C(41)-C(4)	120.0(3)
C(46)-C(41)-C(4)	123.3(3)
C(41)-C(42)-C(43)	121.8(3)
C(44)-C(43)-C(42)	119.9(3)
C(45)-C(44)-C(43)	119.3(3)
C(44)-C(45)-C(46)	120.5(3)
C(45)-C(46)-C(41)	121.7(3)
C(51)-C(5)-C(6)	113.1(3)
C(51)-C(5)-C(4)	114.4(2)
C(6)-C(5)-C(4)	108.2(3)
C(52)-C(51)-C(5)	115.1(3)
N(1)-C(6)-C(5)	110.7(3)
N(1)-C(12)-C(14)	115.4(2)

N(1)-C(12)-C(13)	107.9(3)
C(14)-C(12)-C(13)	113.7(3)
O(13)-C(13)-O(13A)	89.8(4)
O(13)-C(13)-C(12)	113.2(4)
O(13A)-C(13)-C(12)	115.9(5)
C(19)-C(14)-C(15)	117.2(3)
C(19)-C(14)-C(12)	122.6(4)
C(15)-C(14)-C(12)	120.2(4)
C(16)-C(15)-C(14)	120.9(4)
C(17)-C(16)-C(15)	120.4(5)
C(16)-C(17)-C(18)	119.7(4)
C(17)-C(18)-C(19)	120.7(5)
C(18)-C(19)-C(14)	121.2(5)

Table 4. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for 1.
 The anisotropic displacement factor exponent takes the form:
 $-2 \pi^2 [h^2 a^*^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
N(1)	42 (1)	53 (2)	51 (2)	7 (2)	1 (1)	-9 (1)
O(13)	64 (3)	97 (4)	52 (3)	-5 (3)	11 (2)	-22 (3)
O(13A)	53 (3)	176 (7)	72 (3)	12 (4)	-6 (2)	-45 (4)
O(31)	90 (4)	49 (3)	95 (4)	-2 (3)	-9 (3)	-1 (3)
O(31A)	52 (4)	55 (5)	89 (5)	16 (4)	4 (4)	-11 (3)
C(2)	44 (2)	49 (2)	50 (2)	1 (2)	2 (1)	-4 (2)
C(3)	44 (2)	50 (2)	41 (2)	0 (2)	6 (1)	-8 (2)
C(31)	57 (2)	55 (2)	61 (2)	-2 (2)	10 (2)	-12 (2)
C(4)	41 (2)	58 (2)	38 (2)	1 (2)	8 (1)	-2 (2)
C(41)	45 (2)	51 (2)	49 (2)	-3 (2)	9 (1)	-4 (2)
C(42)	50 (2)	90 (3)	54 (2)	8 (2)	1 (1)	3 (2)
C(43)	47 (2)	118 (4)	85 (3)	16 (3)	-1 (2)	3 (2)
C(44)	44 (2)	88 (3)	110 (3)	7 (3)	19 (2)	3 (2)
C(45)	55 (2)	78 (3)	78 (2)	-5 (2)	26 (2)	-9 (2)
C(46)	49 (2)	69 (3)	56 (2)	-5 (2)	9 (1)	-4 (2)
C(5)	50 (2)	47 (2)	58 (2)	11 (2)	7 (1)	-2 (2)
C(51)	66 (2)	49 (2)	75 (2)	-6 (2)	16 (2)	-7 (2)
C(52)	143 (5)	68 (3)	140 (4)	-29 (3)	40 (4)	-26 (3)
C(6)	47 (2)	56 (2)	57 (2)	12 (2)	5 (1)	-9 (2)
C(12)	39 (2)	63 (2)	65 (2)	12 (2)	-8 (1)	-9 (2)
C(13)	64 (2)	92 (3)	80 (3)	7 (3)	-13 (2)	-24 (2)
C(14)	32 (1)	75 (3)	68 (2)	7 (2)	-1 (1)	-9 (2)
C(15)	64 (3)	90 (3)	102 (3)	9 (3)	16 (2)	3 (2)
C(16)	73 (3)	115 (4)	108 (4)	-16 (4)	28 (3)	17 (3)
C(17)	51 (2)	145 (5)	81 (3)	12 (4)	12 (2)	-8 (3)
C(18)	56 (2)	126 (4)	79 (3)	20 (3)	9 (2)	-18 (3)
C(19)	46 (2)	73 (3)	78 (2)	10 (2)	8 (2)	-16 (2)